

Generation of Human Aldose Reductase Mutants of Cys298

by

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Generation of Human Aldose Reductase Mutants of Cys298

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## Thesis Abstract

Human aldose reductase (hAR) uses the oxidation of NADPH to NADP<sup>+</sup> in order to catalyze the reduction of aldehydes to their corresponding alcohols. This enzyme is important because of its role in the polyol pathway and hyperglycemia-induced tissue damage. Several studies have shown that Cys298 has a role in the activation of hAR to its oxidized form. Even so, though it is not a catalytic residue the mechanism of hAR activation is not yet clearly understood. Structure, function and kinetic studies of Cys298 mutants will further our understanding of the role of this residue in substrate binding and interaction. This thesis project forms the foundation for future investigations since it provides an efficient, reproducible and reliable protein expression and protein purification system for three Cys298 mutants: Cys298Ala, Cys298Asp, and Cys298Gly.

## **Acknowledgements**

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## Table of Contents

Title Page.....	i
Signature Page.....	ii
Copyright Page.....	iii
Abstract.....	iv
Acknowledgments.....	v
Table of Contents.....	vi-viii
List of Tables.....	ix
List of Figures.....	x-xi
List of Symbols and Abbreviations.....	xii-xiii
Chapter I	
Introduction.....	1-23
Aldo-Keto Reductases.....	1-2
Natural Substrates.....	3-9
Polyol Pathway.....	10-11
Human Aldose Reductase and Hyperglycemic Tissue Damage.....	11-12
Kinetic Properties of Native and Activated hAR.....	12-14
A Specific Post-Translational Modification.....	14-17
Hypotheses on Cys298 Oxidation.....	17-18
Previous Work: Kinetic Studies.....	19-23
Chapter II	
Materials and Methods.....	23-37
Materials.....	23-24
Strains, Media and Buffers.....	24-25
pET-15b.....	25-26

	pWhitescript.....	27
	pUC18.....	27-28
	Polymerase Chain Reaction.....	29-31
	Preparation of -80°C Bacterial Glycerol Stock.....	31-32
	Preparation of Competent Cells with Calcium Chloride Treatment .....	32
	XL1-Blue <i>E. coli</i> Transformation.....	32-33
	Isolation of Plasmid DNA (QIAprep ® Spin Miniprep Kit).....	33-34
	DNA Sequencing.....	34
	BL21(DE3) <i>E. coli</i> Transformation.....	35
	SDS-PAGE.....	35-37
Chapter III	Protein Expression.....	38
Chapter IV	Protein Purification.....	38-40
	Sonication.....	38-39
	Immobilized Metal Affinity Chromatography.....	39
	Dialysis.....	39
	Protein Quantification: Bradford Reagent Assay.....	40
	Thrombin Cleavage of Histidine Tag.....	40
Chapter V	Results.....	41-83
	DNA Sequencing.....	41-77
	Protein Expression.....	78-79
	Immobilized Metal Affinity Chromatography.....	80-81
	Dialysis and Thrombin Cleavage of Histidine Tag.....	82-83
Chapter VII	Conclusion.....	84-85

Chapter VII References.....86-89



## List of Tables

2.1	Mutagenesis Control Reaction.....	30
2.2	Mutant Reaction.....	30
2.3	Thermal Cycling Parameters.....	31
2.4	Separating Gel (10 mL).....	36
2.5	Stacking Gel (5 mL).....	36

## List of Figures

1.1	Phylogenetic Tree of Human Aldo-Keto Reductases.....	2
1.2	Lipid Peroxidation Products.....	4
1.3	Lipid Peroxidation Products.....	5
1.4	Formation of Advanced Glycation End-Products.....	7
1.5	Lysine-derived AGEs.....	8
1.6	Arginine-derived AGEs.....	9
1.7	Polyol Pathway.....	10
1.8	Kinetic Properties of Native and Activated Human Aldose Reductase.....	13
1.9	Cysteine and Serine Amino Acids.....	15
1.10	Kinetic Studies of Cysteine Mutants with Different Substrates.....	16
1.11	Oxidative Post-Translational Modifications to Cysteine in Proteins.....	18
1.12	Kinetic Study of C298S Mutant with Different Substrates.....	20
1.13	Omit Map of Tyr209 in wild-type and C298S hAR.....	22
2.1	Map of pET-15b Vector.....	26
2.2	Map of pUC18.....	28
5.1	C298A Tube 1 Forward Sequence.....	42
5.2	C298A Forward DNA Chromatogram.....	43-47
5.3	C298A Tube 1 Reverse Sequence.....	48
5.4	C298A Reverse DNA Chromatogram.....	49-53
5.5	C298D Tube 2 Forward Sequence.....	54
5.6	C298D Forward DNA Chromatogram.....	55-59
5.7	C298D Tube 2 Reverse Sequence.....	60

5.8	C298D Reverse DNA Chromatogram.....	61-65
5.9	C298G Tube 1 Forward Sequence.....	66
5.10	C298G Forward DNA Chromatogram.....	67-71
5.11	C298G Tube 1 Reverse Sequence.....	72
5.12	C298G Reverse DNA Chromatogram.....	73-77
5.13	Uninduced/Induced Samples of C298G.....	79
5.14	IMAC Column Samples of C298A.....	81
5.15	Dialysis and Thrombin Digestion Samples of C298D.....	83

## List of Symbols and Abbreviations:

%	percent
®	registered
µg	microgram
µL	microliter
µM	micromolar
3-DG	3-deoxyglucosone
A	adenosine
Abs	UV-Vis absorbance
AGEs	advanced glycation end-products
AKRs	aldo-keto reductases
Ala	alanine
Amp	ampicillin
Asp	aspartate
BME	β-mercaptoethanol
Bp	base pair
C	cytosine
C298A	cysteine to alanine mutant
C298D	cysteine to aspartate mutant
C298G	cysteine to glycine mutant
CEL	N <sup>ε</sup> -(carboxyethyl)-lysine
CMA	N <sup>ε</sup> -(carboxymethyl)-arginine
CML	N <sup>ε</sup> -(carboxymethyl)-lysine
Cys	cysteine
ddH <sub>2</sub> O	double-distilled H <sub>2</sub> O
DNA	deoxyribonucleic acid
dsDNA	double stranded DNA
Ft	foot
G	guanine
GA	glycoaldehyde
Gly	glycine
GOLD	glyoxal-lysine dimer
GSH	tripeptide glutathione (Gly-Cys-Gly)
GSNO	S-nitrosoglutathione
GSSG	glutathione
hAR	human aldose reductase
HHE	4-hydroxy-2(E)-hexenal
HNE	4-hydroxy-2(E)-nonenal
HPHE	4-hydroperoxy-2(E)-hexenal
HPNE	4-hydroperoxy-2(E)-nonenal
IgG	bovine γ-globulin (immunoglobulin G)
IMAC	immobilized metal affinity chromatography
IPTG	isopropyl-1-thio-β-D-galactopyranoside
K	kilo

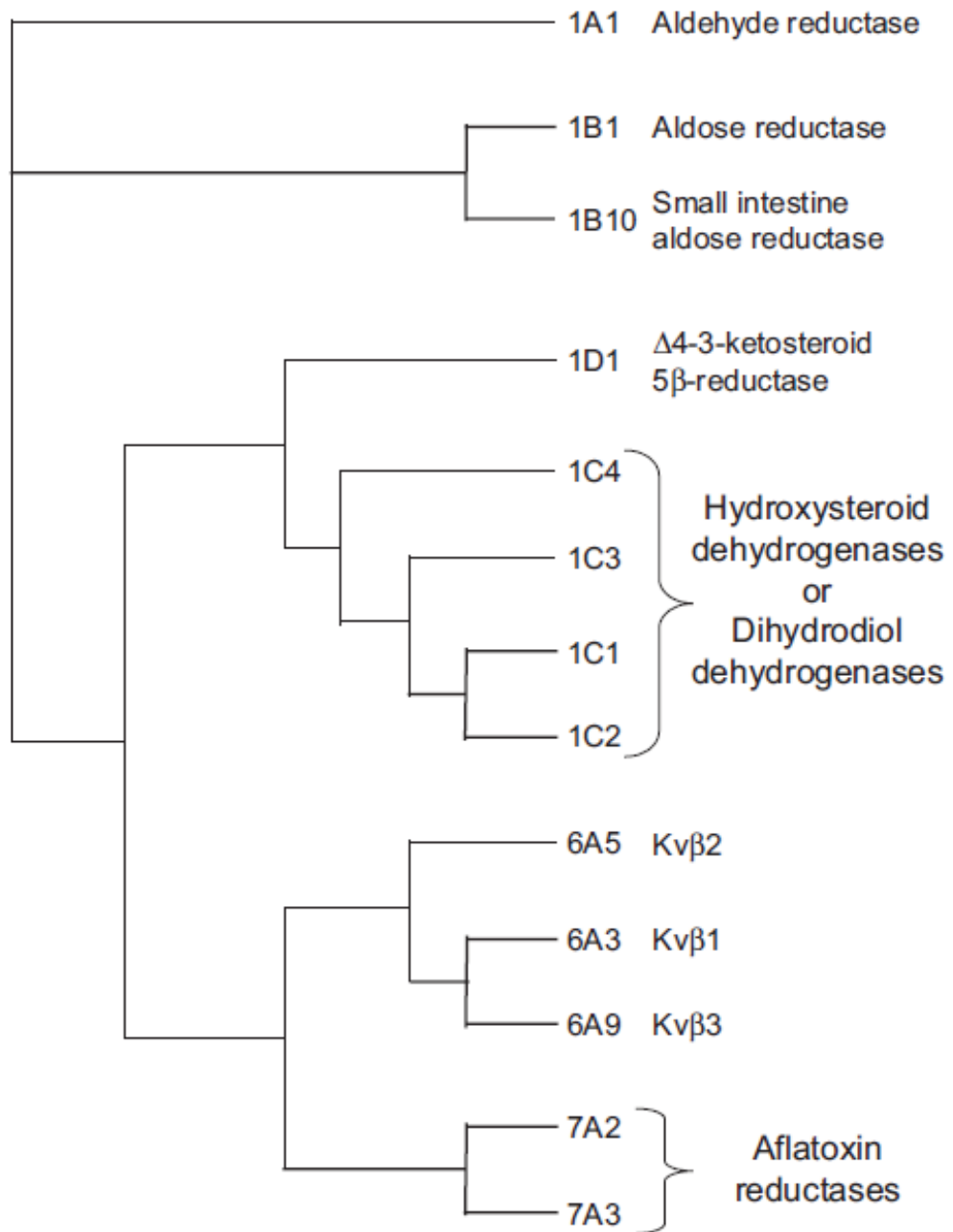
$k_{cat}$	turnover number
$k_{cat}/K_m$	catalytic efficiency
$K_i$	inhibition constant
$K_m$	Michaelis constant
LB	Luria-Broth
M	molar
MDA	malonaldehyde
MES	4-morpholineethanesulfonic acid
mg	milligram
mL	milliliter
mM	milimolar
MOLD	methylglyoxal-lysine dimer
MWCO	molecular weight cut-off
Ng	nanogram
OHE	4-oxo-2(E)-hexenal
ONE	4-oxo-2(E)-nonenal
PAGE	polyacrylamide gel electrophoresis
PBS	phosphate buffered saline
PCR	polymerase chain reaction
POVPC	1-palmitoyl-2-(5-oxo- <i>valeroyl</i> )-sn-glycero-3-phosphocholine
rpm	revolutions per minute
S	second
SDS	sodium dodecyl sulfate
SOC	super optimal broth with catabolite repression
T	thymidine
TEMED	tetramethylethylenediamine
$T_m$	melting temperature
U	unit
V	volts
v/v	volume per volume
$V_{max}$	maximum velocity
w/v	weight per volume
X-Gal	5-bromo-4-chloro-3-indoyl- $\beta$ -D-galactopyranoside

## **Chapter I: Introduction**

### **I. Aldo-Keto Reductases**

Human aldose reductase (hAR) belongs to a group of structurally similar proteins called aldo-keto reductases (AKRs). All plants, animals and fungi utilize AKRs to catalyze the reduction of aldehydes and ketones to primary or secondary alcohols.<sup>1</sup> These AKRs play an important role in the detoxification of toxic aldehydes and ketones that are produced endogenously (within the cell or organism) or exogenously (outside the cell or organism).<sup>1</sup> Humans express 13 AKR proteins: aldehyde reductase (AKR1A1), aldose reductases (AKR1B1, AKR1B10), hydroxysteroid dehydrogenases (AKR1C1, AKR1C2, AKR1C3, AKR1C4),  $\Delta^4$ -3-ketosteroid-5- $\beta$ -reductase (AKR1D1), Kv $\beta$  voltage-gated potassium channel proteins (AKR6A3, AKR6A5, AKR6A9), and aflatoxin reductases (AKR7A2, AKR7A3).<sup>1</sup> Below is the phylogenetic tree of human aldo-keto reductases.

Figure 1.1: Phylogenetic Tree of Human Aldo-Keto Reductases<sup>1</sup>



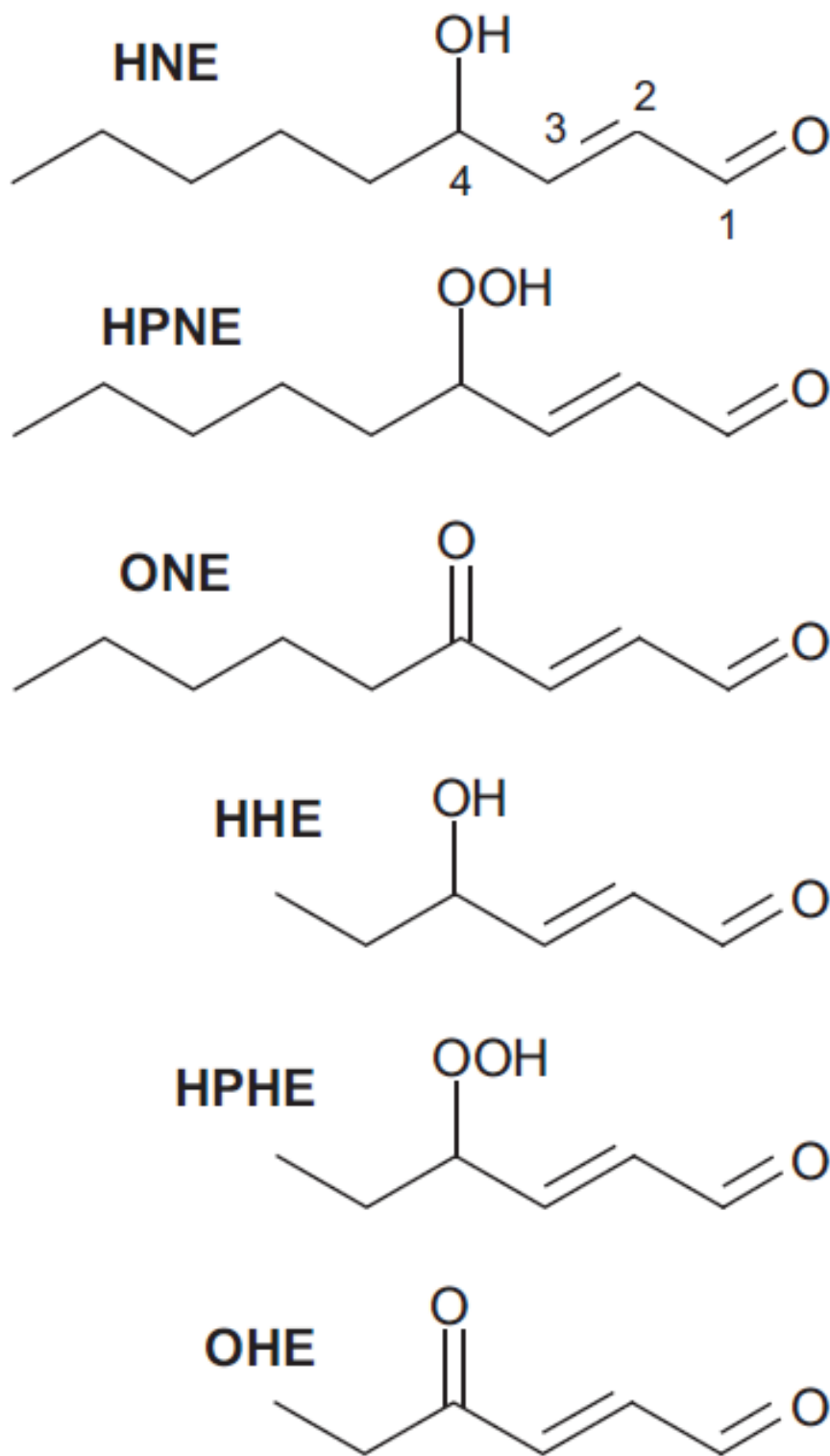
## II. Natural Substrates

For aldose reductases like hAR (AKR1B1), the tight binding to NADPH predominantly provides the energy for catalysis.<sup>1</sup> This is why a wide range of aldehydes can be efficiently reduced even when there is little substrate affinity. Human aldose reductase's natural substrates include lipid peroxidation products and advanced glycation end-products (AGEs).<sup>1</sup> Lipid peroxidation is the oxidative degradation of lipids to yield aldehydes such as methylglyoxal, 3-deoxyglucosone (3-DG), 4-hydroxy-2(E)-nonenal (HNE) and 1-palmitoyl-2-(5-oxovaleroyl)-sn-glycerol-3-phosphocholine (POVPC).<sup>1</sup> Chemical structures of various lipid peroxidation products are shown in the following figures (Figure 1.2 and Figure 1.3).





Figure 1.3: Lipid Peroxidation Products<sup>2</sup>



Meanwhile, AGEs are modifications of proteins or lipids that are produced when proteins or lipids react with aldose sugars through non-enzymatic glycosylation.<sup>3</sup> The figure below (Figure 1.4) illustrates the process by which an aldehyde (glucose) reacts with a protein to form a Schiff base (aldimine), undergoes Amadori rearrangement to become a more stable ketoamine and then is oxidatively cleaved by reactive nitrogen or oxygen species to produce an AGE.<sup>1</sup> Figure 1.5 and Figure 1.6 show both lysine-derived and arginine-derived AGEs.

Figure 1.4: Formation of Advanced Glycation End Products<sup>1</sup>

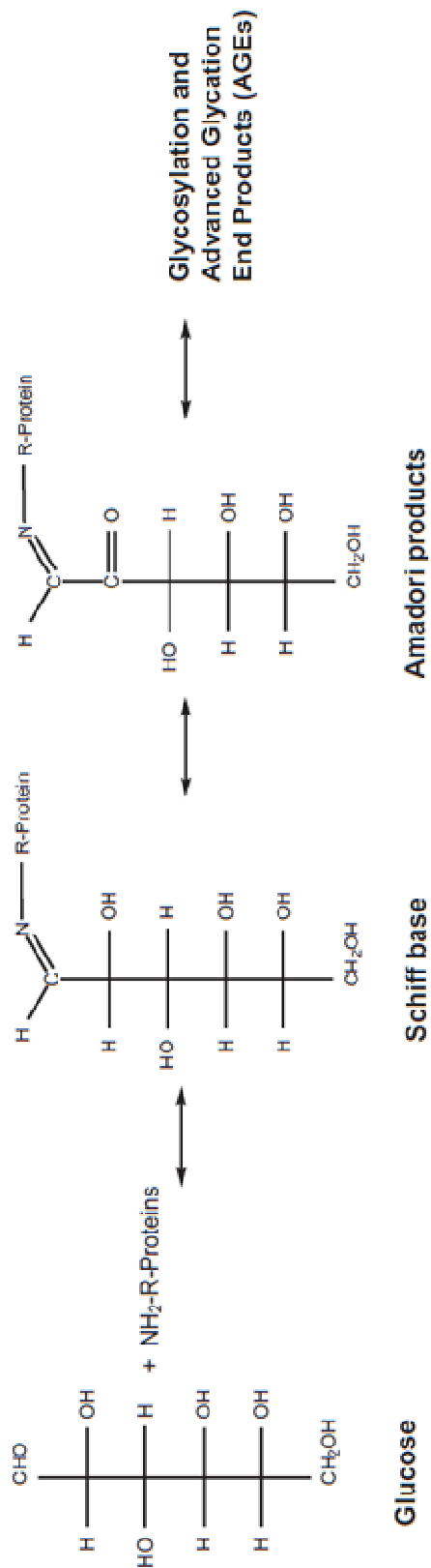


Figure 1.5: Lysine Derived AGEs<sup>4</sup>

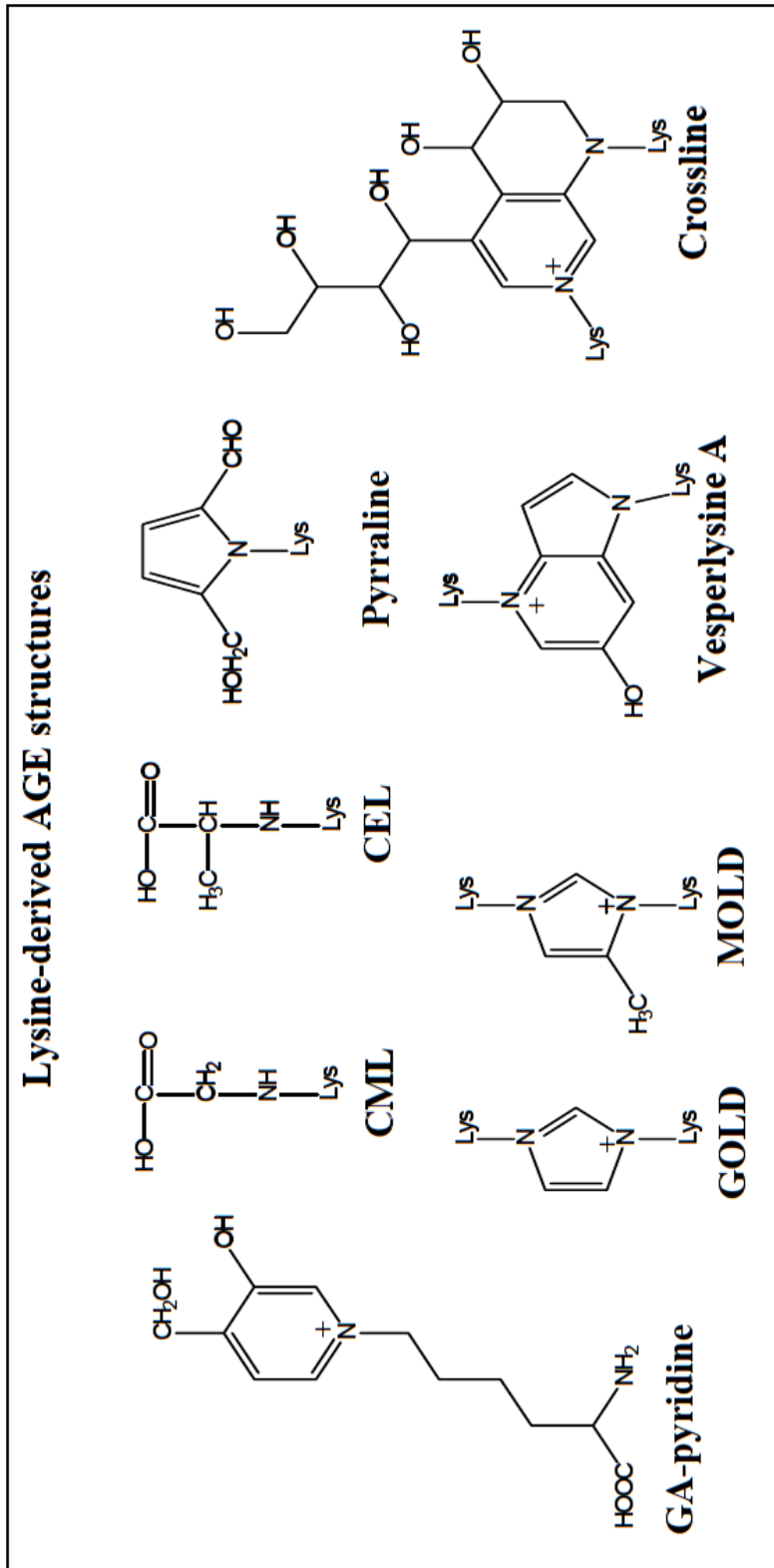
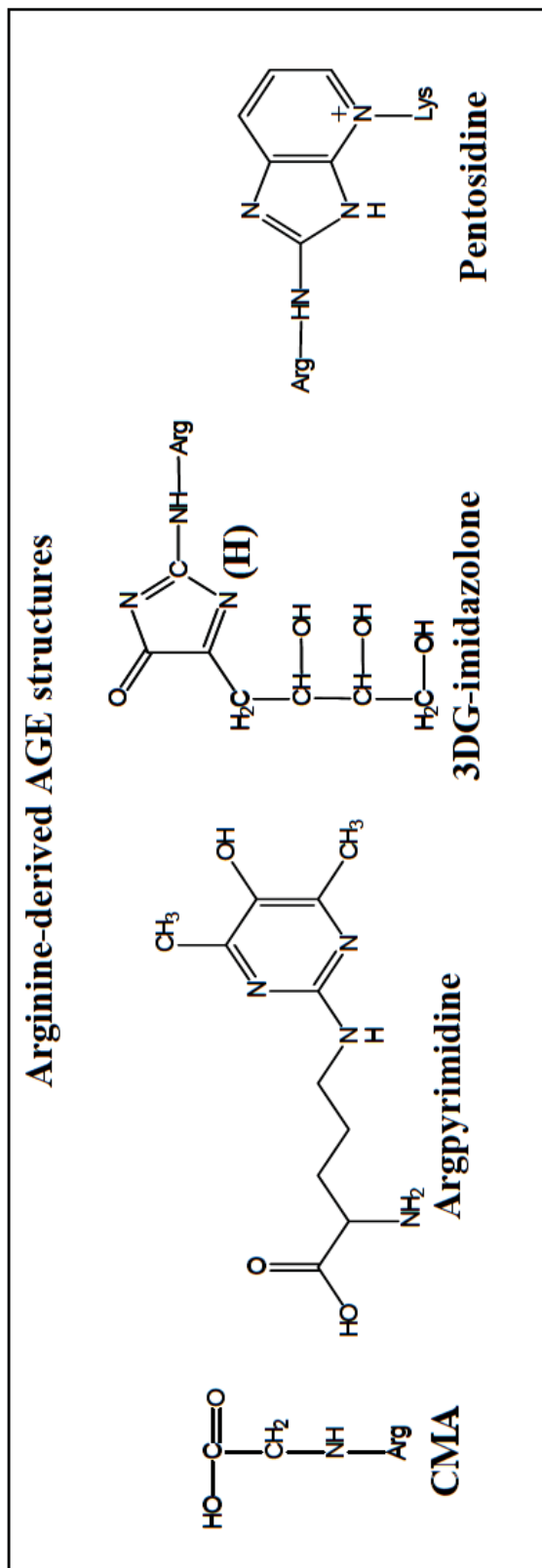


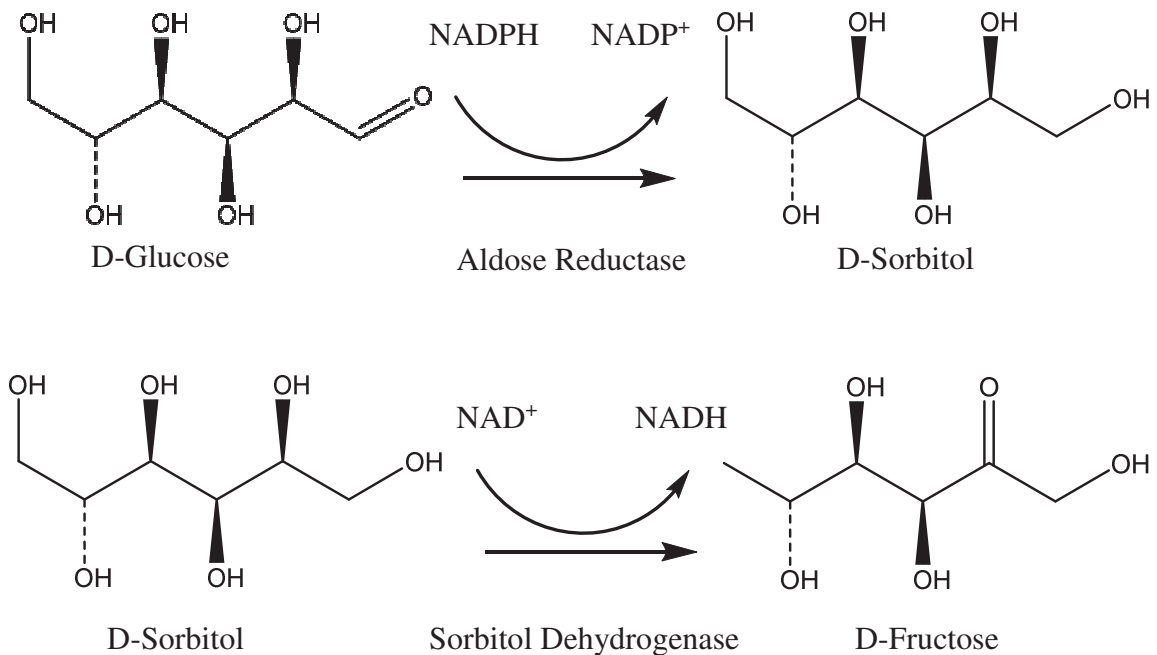
Figure 1.6: Arginine-derived AGEs<sup>4</sup>



### III. Polyol Pathway

Normally, humans use two important pathways for metabolizing excess glucose. Glycogenesis is for short-term energy storage and through this pathway the glucose is stored as glycogen. Meanwhile, triacylglycerol biosynthesis is for long-term energy storage and through this pathway the glucose is converted to triacylglycerols (fatty acids). However, when the human body is overwhelmed with excess glucose (hyperglycemia) it utilizes an additional pathway to metabolize glucose.<sup>5</sup> This supplementary pathway is the polyol pathway (Figure 1.7).<sup>5</sup> During normoglycemia (normal glucose levels) less than 3% of glucose is metabolized through this pathway, but during hyperglycemia as much as 30% of the total glucose is metabolized through this pathway.<sup>6,7</sup>

Figure 1.7: Polyol Pathway



Aldose reductase is an important enzyme in the polyol pathway because it catalyzes the rate-determining step, which is the first step in this pathway. Aldose reductase uses the oxidation of NADPH to NADP<sup>+</sup> to catalyze the reduction of aldose aldehydes to their corresponding alcohols.<sup>8</sup>

#### **IV. Human Aldose Reductase and Hyperglycemic Tissue Damage**

Even with the assistance of the polyol pathway the body cannot keep pace with the excess glucose. When this situation occurs tissues are damaged and diabetic complications ensue.<sup>5,6,7</sup> The “role of [human] aldose reductase in mediating hyperglycemic injury remains obscure,” but there are several hypotheses for how it is involved in tissue damage.<sup>7</sup> One idea is that tissue damage is caused by the buildup of sorbitol from increased hAR activity. Sorbitol is impermeable to the cell membrane and when it accumulates inside the cell it causes oxidative stress, protein insolubilization, glycation and diabetic complications.<sup>7</sup> Moreover, sorbitol buildup leads to increased osmotic pressure in the kidneys because of its role in balancing the osmotic pressure of extracellular NaCl during urine concentration.<sup>6</sup> Another theory is that tissue damage is caused by oxidative stress induced by the depletion of NADPH and NAD<sup>+</sup> from increased hAR and sorbitol dehydrogenase activity.<sup>6</sup> Utilization of the polyol pathway depletes NADPH and NAD<sup>+</sup>, which results in “metabolic imbalances” in “tissues that undergo insulin-independent uptake of glucose.”<sup>6</sup> Glucose transport proteins 1, 2, 3 and 5 (SLC2A1, SLC2A2, SLC2A3, SLC2A5) are insulin-independent and transport glucose across various tissues of the body such as the eyes, nerves and kidneys.<sup>9</sup> So with diminished quantities of these important cofactors (NADPH and NAD<sup>+</sup>) less glucose is transported to various areas of the body. This is why diabetes is often associated with



damage in the ocular lens, retina, peripheral nervous system, and renal glomerulus.<sup>6</sup> Additionally, NADPH depletion “compromises antioxidative defenses.”<sup>7</sup> Glutathione reductase uses NADPH to reduce glutathione disulfide to glutathione.<sup>7,10</sup> Glutathione is important for resisting oxidative stress, maintaining the reducing environment of the cell, and is a major reducing agent for disulfides and peroxides.<sup>10</sup> Clearly, hAR does have a role in causing hyperglycemic tissue damage and inhibitors of hAR have been proven to prevent, delay and reverse tissue damage due to diabetic complications.<sup>7</sup>

#### **V. Kinetic Properties of Native and Activated hAR**

At normal glucose levels, human aldose reductase (hAR) exists mainly in its native (unoxidized) form.<sup>8</sup> During hyperglycemia, native human aldose reductase is oxidized to its activated, inhibitor-resistant form.<sup>8</sup>

In a study done by Grimshaw et al. (1996), they examined the kinetic properties of naturally occurring activated and native human placental aldose reductase. They observed that with DL-glyceraldehyde as substrate the activated hAR had a significantly higher Michaelis constant ( $K_m$ ), and higher maximum velocity ( $V_{max}$ ) than that of the native hAR (Figure 1.8).<sup>8</sup> Also with sorbinil as the inhibitor the activated hAR had a higher inhibition constant ( $K_i$ ) (Figure 1.8).<sup>8</sup>

Figure 1.8: Kinetic Properties of Native and Activated Human Aldose Reductase<sup>8</sup>

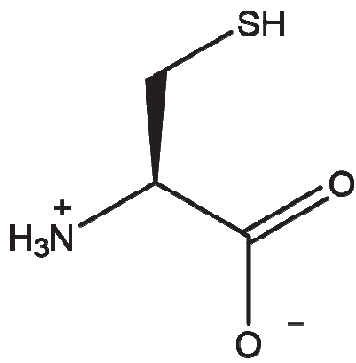
Parameter <sup>a</sup>	Recombinant enzyme		Human placenta	
	Native	Oxidized	Native <sup>b</sup>	Oxidized <sup>c</sup>
$V$ ( $s^{-1}$ )	0.45	0.50	0.62	1.5
$K$ (mM)	0.047	2.4	0.024	6.8
$V/K$ ( $mM^{-1} s^{-1}$ )	9.6	0.21	26	0.22
$K_i$ Sorbinil ( $\mu M$ )	0.5	380	0.5	400

The higher  $V_{\max}$  of activated hAR indicates that it can catalyze the corresponding forward reduction reaction at a faster maximum rate than native hAR. The higher  $K_m$  of activated hAR indicates that it has a lower affinity for the substrate.<sup>8</sup> Sorbinil is a non-competitive inhibitor for the forward reduction reaction and it interacts with the enzyme regardless of whether substrate is bound to it. The higher  $K_i$  for the activated hAR shows that a higher concentration of sorbinil was required to decrease the  $V_{\max}$  by one-half.<sup>8</sup> So clearly, activated hAR has a detrimental effect on inhibitor potency and enzyme-substrate interaction.

#### **VI. A Specific Post-Translational Modification**

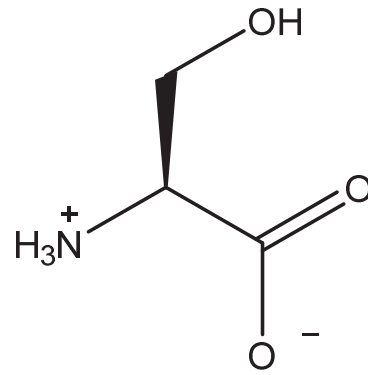
The work done by Petrash et al. (1992) located a specific residue that contributes to hAR activation *in vitro*. Petrash et al. (1992) carried out kinetic studies of recombinant C80S, C298S, and C303S mutants to determine which residue modifications likely contributes to the activation of hAR.<sup>11</sup> These three residues were selected as potential candidates because they are located in the vicinity of the active site and prone to oxidation.<sup>11</sup> Also, all of these residues were modified from a cysteine (thiol side chain) to a serine (hydroxyl side chain) (Figure 1.9).

Figure 1.9: Cysteine and Serine Amino Acids<sup>13</sup>



Cysteine

side chain pKa: 8



Serine

side chain pKa: 13

Figure 1.10: Kinetic Studies of Cysteine Mutants with Different Substrates<sup>11</sup>

*Kinetic constants for wild type and mutant aldose reductases*

Values of kinetic constants are mean  $\pm$  S.E. ( $n = 15$ ). The indicated substrates were varied over a concentration range of at least 0.1–2 times  $K'_m$ .  $K'_m$  values are apparent Michaelis constants for the indicated substrate.

	Wild type	C80S	C298S	C303S
<b>D-Xylose</b>				
$K'_m$ (mM)	10.2 $\pm$ 1.49	15.6 $\pm$ 1.55	543 $\pm$ 102 <sup>a</sup>	290 $\pm$ 36 <sup>a</sup>
$k_{cat}$ (s <sup>-1</sup> )	1.24 $\pm$ 0.06	1.52 $\pm$ 0.04	8.48 $\pm$ 0.85 <sup>a</sup>	0.81 $\pm$ 0.4
$k_{cat}/K'_m$ (s <sup>-1</sup> M <sup>-1</sup> )	122 $\pm$ 13.5	97.6 $\pm$ 8.07	15.7 $\pm$ 1.52 <sup>a</sup>	2.77 $\pm$ 0.23 <sup>a</sup>
<b>D-Glucose</b>				
$K'_m$ (mM)	212 $\pm$ 26.7	196 $\pm$ 24	>1000 <sup>a</sup>	881 $\pm$ 277 <sup>a</sup>
$k_{cat}$ (s <sup>-1</sup> )	0.086 $\pm$ 0.004	0.359 $\pm$ 0.026 <sup>a</sup>	>1.95 <sup>a</sup>	0.037 $\pm$ 0.004 <sup>a</sup>
$k_{cat}/K'_m$ (s <sup>-1</sup> M <sup>-1</sup> )	0.407 $\pm$ 0.033	1.87 $\pm$ 0.269 <sup>a</sup>	ND <sup>b</sup>	0.085 $\pm$ 0.010 <sup>a</sup>
<b>p-Nitrobenzaldehyde</b>				
$K'_m$ ( $\mu$ M)	15.7 $\pm$ 1.70	34.6 $\pm$ 6.79 <sup>a</sup>	202 $\pm$ 14.1 <sup>a</sup>	254 $\pm$ 20.2 <sup>a</sup>
$k_{cat}$ (s <sup>-1</sup> )	0.469 $\pm$ 0.011	0.377 $\pm$ 0.014	3.33 $\pm$ 0.092 <sup>a</sup>	0.486 $\pm$ 0.016
$k_{cat}/K'_m$ (s <sup>-1</sup> M <sup>-1</sup> )	29,738 $\pm$ 2,689	10,865 $\pm$ 1,881 <sup>a</sup>	16,432 $\pm$ 741 <sup>a</sup>	1,912 $\pm$ 94 <sup>a</sup>
<b>Benzaldehyde</b>				
$K'_m$ ( $\mu$ M)	72.8 $\pm$ 11.3	27.6 $\pm$ 3.49 <sup>a</sup>	1,061 $\pm$ 208 <sup>a</sup>	67.7 $\pm$ 9.39
$k_{cat}$ (s <sup>-1</sup> )	0.443 $\pm$ 0.044	0.758 $\pm$ 0.018 <sup>a</sup>	4.62 $\pm$ 0.121 <sup>a</sup>	0.299 $\pm$ 0.011 <sup>a</sup>
$k_{cat}/K'_m$ (s <sup>-1</sup> M <sup>-1</sup> )	6,093 $\pm$ 742	27,473 $\pm$ 3,033 <sup>a</sup>	4,404 $\pm$ 377 <sup>a</sup>	4,420 $\pm$ 501

<sup>a</sup>  $p < 0.001$  compared to ALR2:wild type.

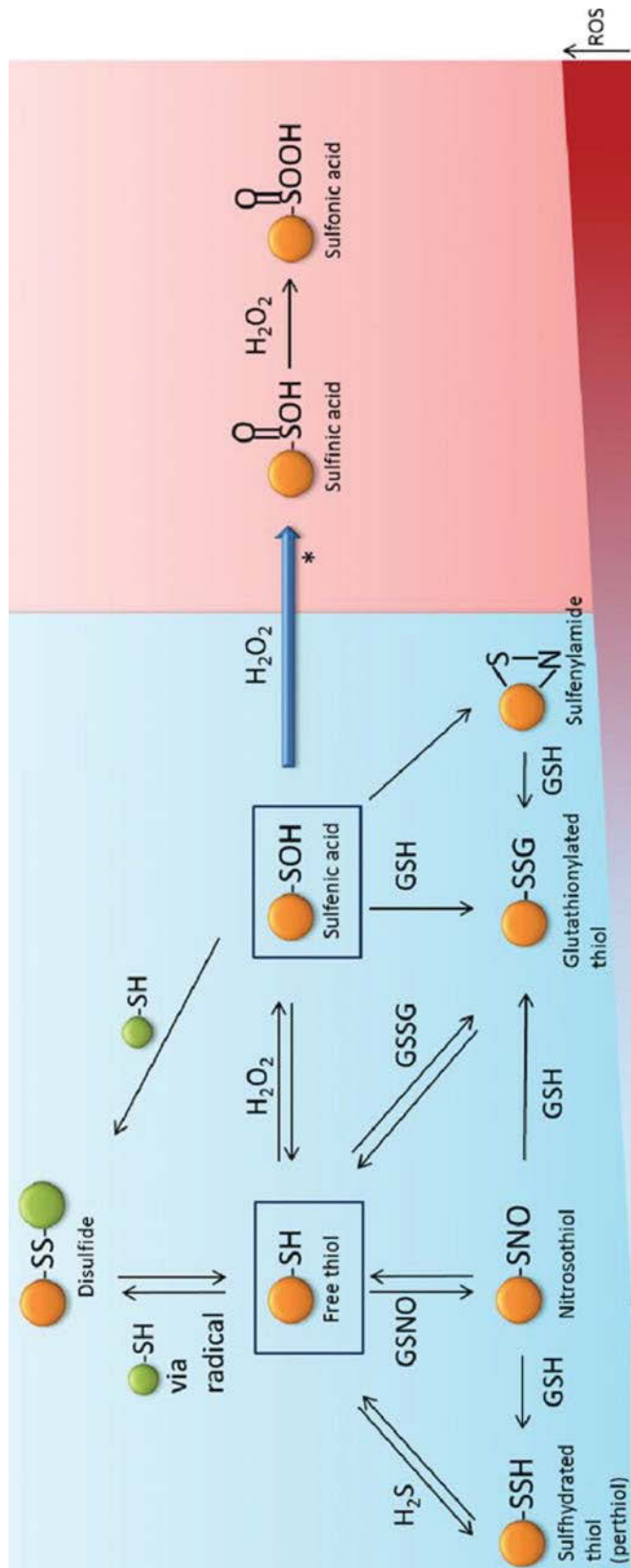
<sup>b</sup> ND, value could not be accurately computed due to uncertainty in  $K'_m$  and  $k_{cat}$  components.

From kinetic studies of these recombinant mutants of hAR with different substrates, the researchers noted that the greatest changes to  $K_m$ , turnover number ( $k_{cat}$ ) and catalytic efficiency ( $k_{cat}/K_m$ ) correspond with the C298S mutant (Figure 1.10).<sup>11</sup> This C298S mutant had a significantly higher  $K_m$ , higher  $k_{cat}$  and lower catalytic efficiency for all the substrates (Figure 1.10).<sup>11</sup> If you focus on the values for C298S with D-glucose as substrate you will notice that its  $K_m$  and  $k_{cat}$  are significantly larger and its catalytic efficiency is so small that it could not be accurately determined.<sup>11</sup> The researchers attributed these unobvious changes to the altered interaction of residue 298 with the nicotinamide ring of NADPH and the changed characteristics of the active site.<sup>11</sup>

## VII. Hypotheses on Cys298 Oxidation

Studies have shown that oxidation of Cys298 prevents the NADPH-binding loop from completely closing.<sup>1,11</sup> This modification allows  $NADP^+$  to be released faster. Also, since  $NADP^+$  release is the rate limiting step in hAR catalysis this alteration leads to increased turnover rate, increased catalysis, and increased inhibition constant.<sup>1,11</sup> Cys298 mutants like C298S model the oxidation of cysteine ( $S^{-2}$ ) to its various oxidation states: sulfenic acid ( $S^0$ ), sulfinic acid ( $S^{+2}$ ) and sulfonic acid ( $S^{+4}$ ).<sup>1</sup> Even so, no one is certain of how exactly Cys298 becomes oxidized, but oxidative post-translational modification of cysteine residues can occur via sulfur-nitrosylation, sulfhydration, sulfur-glutathionylation, disulfide bond formation, sulfenylation, and oxidation to sulfur acids.<sup>1</sup>  
<sup>12</sup> These oxidative post-translational modifications are often induced by reactive nitrogen and oxygen species and glutathiolated and sulfenic acid forms of hAR have been discovered *in vivo*.<sup>1, 12</sup> Figure 1.11 shows common oxidative post-translational modifications that occur to cysteine residues in other proteins.

Figure 1.11: Oxidative Post-Translational Modifications to Cysteine in Proteins<sup>12</sup>



### VIII. Previous Work: Kinetic Studies

The work done by Dr. Balendiran and Richard Cuckovich was published in 2011 and further confirmed that Cys298 plays a crucial role in the activation of hAR. They conducted kinetic studies on the hAR C298S mutant with four different substrates: DL-glyceraldehyde, HNE, D-glucose, and benzyl alcohol (Figure 1.12). For all the substrates the C298S mutant has a significantly larger  $K_m$  than that of the wild-type.<sup>13</sup> The  $k_{cat}$  of the C298S mutant was remarkably higher for only DL-glyceraldehyde and benzyl alcohol; so the mutant required more time to turnover one molecule of those substrates.  $k_{cat}/K_m$  is a measure of the catalytic efficiency of the enzyme and for the mutant this value was lower for all the substrates.<sup>13</sup> Also, benzyl alcohol was the only substrate studied for the reverse oxidation reaction and it is interesting that the C298S mutant had a lower affinity for it when compared to that of the wild-type.<sup>13</sup>



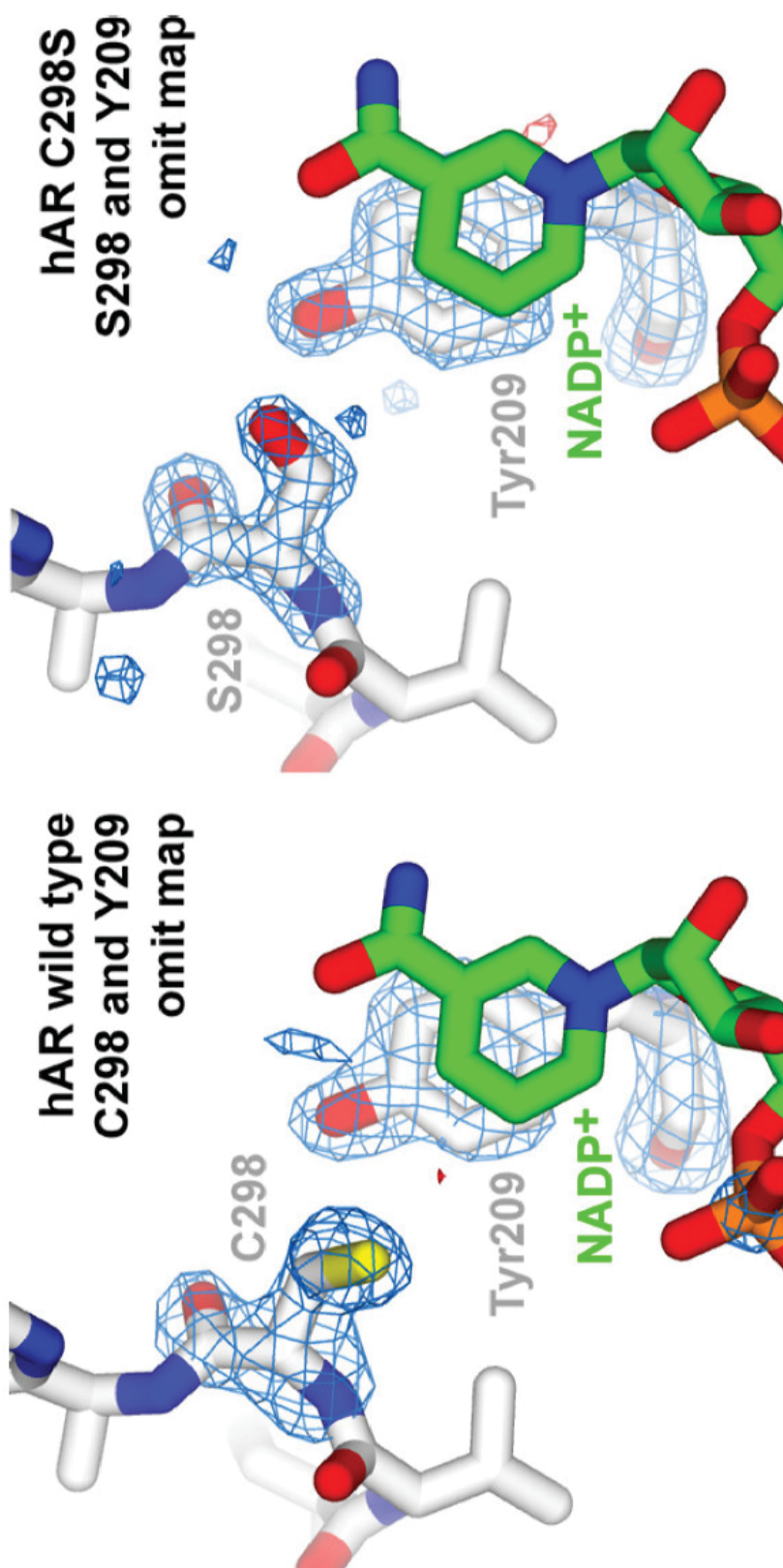
Figure 1.12: Kinetic Study of C298S Mutant with Different Substrates<sup>13</sup>

**Comparison of catalytic activity of wild-type and C298S mutant hAR for carbonyl reduction and alcohol oxidation reactions at 25 °C**  
 The uncertainties are S.D. of the average value.

hAR Protein	Substrate	$K_m$	$k_{cat}$ $s^{-1}$	$k_{cat}/K_m$
Wild-type C298S	DL-Glyceraldehyde	$0.1 \pm 0.01$ mM	$1.55 \pm 0.01$	$14.5$ mM <sup>-1</sup> s <sup>-1</sup>
Wild-type C298S	D-Glucose	$1.2 \pm 0.05$ mM	$3.72 \pm 0.3$	$3.10$ mM <sup>-1</sup> s <sup>-1</sup>
Wild-type C298S	HNE	$55.0 \pm 6.0$ mM	$0.64 \pm 0.05$	$0.011$ mM <sup>-1</sup> s <sup>-1</sup>
Wild-type C298S	Benzyl alcohol	$800 \pm 50.0$ mM	$0.42 \pm 0.04$	$0.0005$ mM <sup>-1</sup> s <sup>-1</sup>
Wild-type C298S		$48.0 \pm 4.0$ $\mu$ M	$0.90 \pm 0.06$	$0.019$ $\mu$ M <sup>-1</sup> s <sup>-1</sup>
Wild-type C298S		$110.0 \pm 8.0$ $\mu$ M	$0.60 \pm 0.02$	$0.006$ $\mu$ M <sup>-1</sup> s <sup>-1</sup>
Wild-type C298S		$1.62 \pm 0.1$ mM	$0.35 \pm .025$	$0.215$ mM <sup>-1</sup> s <sup>-1</sup>
Wild-type C298S		$4.55 \pm 0.3$ mM	$0.42 \pm .02$	$0.092$ mM <sup>-1</sup> s <sup>-1</sup>

From this table (Figure 1.12) Balendiran et al. (2011) concluded that the affinity of the substrate-enzyme complexes for both the forward and reverse reactions are weakened in the C298S mutant.<sup>13</sup> Furthermore, since the mutant had similar trends for  $K_m$  and catalytic efficiency that had been observed by other researchers for activated hAR, the authors concluded that Cys298 was oxidized during the activation of hAR.<sup>7,8,11,13</sup> Moreover, from crystal structures of wild-type and C298S mutant hAR Balendiran et al. (2011) observed that in the wild-type the Cys298 side chain is near the nicotinamide ring of NADPH and is surrounded by hydrophobic residues: Trp20, Trp209, and Trp219.<sup>13</sup> Meanwhile, in the C298S mutant the Ser298 side chain is further away from those hydrophobic residues and forms a hydrogen bond with the phenol side chain of Tyr209.<sup>13</sup> Figure 1.13 shows the interactions between Tyr209 and the wild-type and C298S mutant.

Figure 1.13: Omit Map of Tyr209 in wild-type and C298S hAR<sup>13</sup>



From this crystal structure map (Figure 1.13) Balendiran et al. (2011) concluded that formation of the hydrogen bond makes residue 298 more rigid and greatly increases the polarity of the NADPH binding pocket.<sup>13</sup> This increase in polarity is then likely responsible for decreasing the strength of the enzyme-NADPH and enzyme-substrate interactions due to the conflicting polarities of the hydrophilic serine side chain with the hydrophobic binding pockets.<sup>13</sup>

## **Chapter II: Materials and Methods**

### **I. Materials**

Difco Yeast Extract UF, Bacto Tryptone, Bacto Agar, and 10 mL BD slip-tip disposable syringes were purchased from BD Biosciences, San Jose, CA; ampicillin sodium salt, 4-morpholineethanesulfonic acid, Trizma base (Sigma 7-9), imidazole, and Luria Broth (Miller) were purchased from Sigma-Aldrich Corporation, St. Louis, MO; Fisherbrand 50 mL graduated polypropylene centrifuge tubes, Falcon 15 mL conical centrifuge tubes, PageRuler Plus Prestained Protein Ladder, sodium chloride, magnesium sulfate, potassium chloride, glucose, and glycerol were purchased from Thermo Fisher Scientific Inc., Waltham, MA; Novagen Thrombin Restriction Grade, Novagen 4x SDS sample buffer, Novagen pET-15b DNA, Millex-HV Syringe Filter Unit (0.45  $\mu$ m), and Millex-GV Syringe Filter Unit (0.22  $\mu$ m) were purchased from EMD Millipore, Darmstadt, Germany; Bio-Rad Protein Assay Dye Reagent Concentrate, Protein Standard I Globulin (bovine  $\gamma$ -globulin), Coomassie Brilliant Blue G-250, sodium dodecyl sulfate, TEMED, Resolving Gel Buffer (1.5 M Tris-HCl pH 8.8), Stacking Gel Buffer (0.5 M Tris-HCl pH 6.8), 10x Tris/Glycine/SDS Electrophoresis Buffer and bromophenol blue were purchased from Bio-Rad Laboratories Inc., Hercules, CA; isopropyl-1-thio- $\beta$ -D-

galactopyranoside was purchased from BioPioneer Inc., San Diego, CA; ammonium persulfate, sodium hydrogen phosphate, potassium dihydrogen phosphate, magnesium chloride, and sodium hydroxide were purchased from Avantor Performance Materials Inc., Center Valley, PA; acrylamide/bisacrylamide liquid (37.5:1) was purchased from Roche Diagnostics Corporation, Indianapolis, IN; methanol (absolute), ethyl alcohol (190 proof), acetic acid, and hydrochloric acid were purchased from Pharmco-Aaper, Toronto, Canada; Spectra/Por 1 dialysis tubing (6-8 k MWCO, 23 mm flat width, 100 ft) was purchased from Spectrum Laboratories Inc., Rancho Dominguez, CA; Talon CellThru (IMAC) resin was purchased from Clontech Laboratories Inc., Mountain View, CA;  $\beta$ -mercaptoethanol was purchased from Amresco Inc., Solon, OH; BL21 (DE3) competent *E. coli* and the QuikChange II Site-Directed Mutagenesis Kit were purchased from Agilent Technologies, Santa Clara, CA; QIAprep Spin Miniprep Kit and TE buffer was purchased from Qiagen, Venlo, Netherlands.

## **II. Strains, Media and Buffers**

The XL1-Blue supercompetent *E. coli* strain was included with the QuikChange II Site-Directed Mutagenesis Kit and it was used in all experiments involving the pWhitescript, pUC18 and pET-15b plasmids. The cells were cultured in Luria Broth (Miller) with ampicillin (1% w/v tryptone, 1% w/v NaCl, 0.5% w/v yeast extract, 100  $\mu$ g/mL ampicillin) [LB-Amp]. Transformants were selected on Luria Broth Agar with ampicillin (Luria Broth (Miller), 0.9% w/v bacto agar, 100  $\mu$ g/mL ampicillin) [LB-Amp plates]. Blue-white screening was done with 40  $\mu$ L of 100 mM IPTG and 40  $\mu$ L of 40 mg/mL X-Gal on LB-Amp plates. The BL21(DE3) competent *E. coli* cells were cultured in 2X YT Media with ampicillin (1.6% w/v bacto tryptone, 1% w/v yeast extract, 0.5%

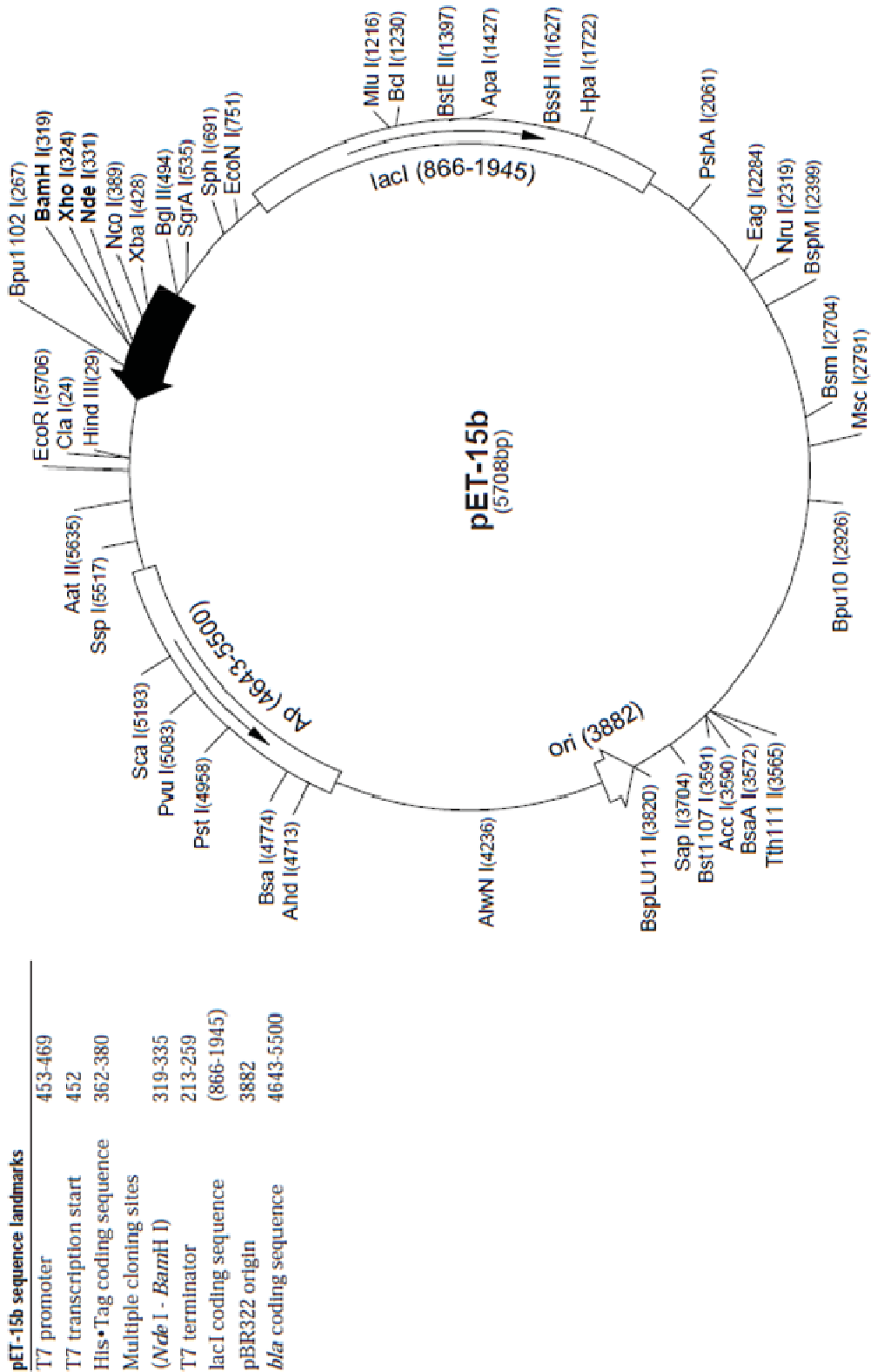
w/v NaCl, pH 7.0, 100 µg/mL ampicillin) and transformants were selected on LB-Amp plates. For protein overexpression, BL21(DE3) transformants were grown in 2X YT media with ampicillin (100 µg/mL) [YT-Amp]. Protein expression was induced in YT-Amp culture with IPTG (1 mM). These following buffers were used for protein purification: extraction buffer (50 mM Tris, 300 mM NaCl, 2 mM BME, pH 7), wash 1 buffer (50 mM Tris, 300 mM NaCl, 2 mM BME, pH 7), wash 2 buffer (10 mM imidazole, 50 mM Tris, 300 mM NaCl, 2 mM BME, pH 7), elution buffer (300 mM imidazole, 50 mM Tris, 300 mM NaCl, 2 mM BME, pH 7), dialysis buffer (50 mM Tris, 300 mM NaCl, 2 mM BME, pH 7).

### **III.pET-15b Vector**

This 5,708 base pair cloning vector (Figure 2.1) contains the ampicillin resistance gene, His-Tag coding sequence, multiple cloning sites and a thrombin cleavage site.

Figure 2.1: Map of pET-15b Vector

Figure adapted from www.chem.agilent.com



#### **IV. pWhitescript**

This 4.5 kbp mutagenesis control plasmid contains the ampicillin resistance gene for antibiotic selection and the *lacZ* gene for blue-white screening. XL1-Blue cells transformed with this plasmid and the oligonucleotide control primers appear blue on LB-Amp plates containing IPTG and X-Gal.

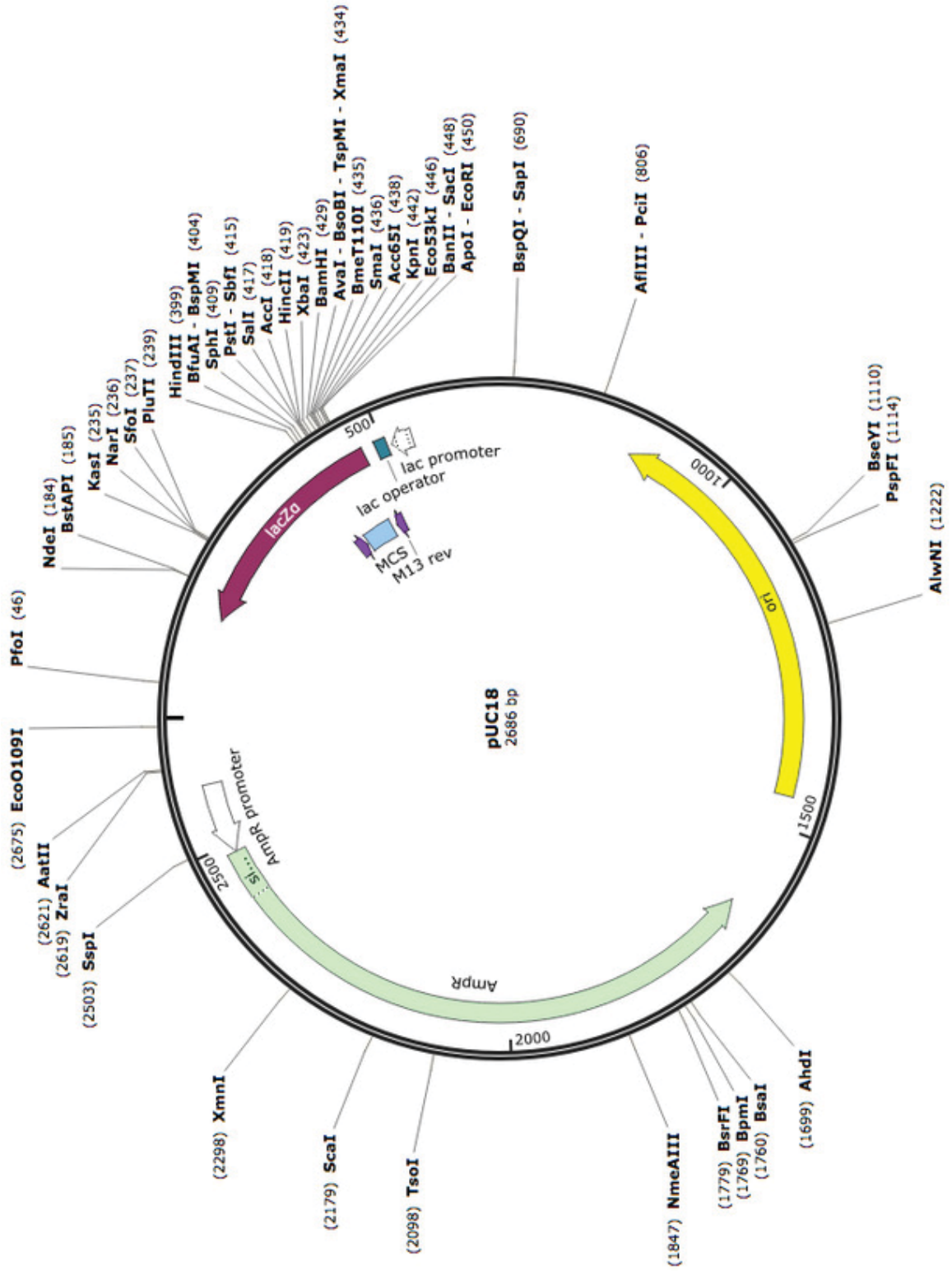
#### **V. pUC18**

This 2,686 bp transformation control plasmid (Figure 2.2) contains the ampicillin resistance gene, *lacZ* gene, and multiple cloning sites. Insertion of the bacterial host's DNA into the multiple cloning site disrupts the *lacZ* gene, resulting in white XL1-Blue cells.



Figure 2.2: Map of pUC18

Figure adapted from [www.snapgene.com](http://www.snapgene.com)



## VI. Polymerase Chain Reaction

A QuikChange II Site-Directed Mutagenesis Kit was used to prepare the mutant and control reactions. Polymerase chain reaction (PCR) was performed in a programmable thermal cycler with the following primers that were synthesized by Eurofins Genomics. Shown below are the primers and underlined are the mutation sites:

C298A Forward: 5' AACAGGAACTGGAGGGTCGCCGCCTTGTTGAGCTGTACC 3'

C298A Reverse: 5' GGTACAGCTCAACAAGGCGGCGACCCTCCAGTTCCTGTT 3'

T<sub>m</sub>: 88.4°C

C298D Forward:

5' CAACAGGAACTGGAGGGTCGACGCCTTGTTGAGCTGTACCT 3'

C298D Reverse:

5' AGGTACAGCTCAACAAGGCGTCGACCCTCCAGTTCCTGTTG 3'

T<sub>m</sub>: 88.0°C

C298G Forward: 5' CAGGAACTGGAGGGTCGGGCCTTGTTGAGCTGTA 3'

C298G Reverse: 5' TACAGCTCAACAAGGCTCCGACCCTCCAGTTCCTG 3'

T<sub>m</sub>: 85.6°C

The mutagenesis control reaction and the mutant reaction was prepared as indicated below in Table 2.1 and Table 2.2:

Table 2.1: Mutagenesis Control Reaction

Volume ( $\mu\text{L}$ )	Reagents
5	10x reaction buffer
2	(5 ng/ $\mu\text{L}$ ) pWhitescript 4.5-kb control plasmid
1.25	(100 ng/ $\mu\text{L}$ ) oligonucleotide control primer #1
1.25	(100 ng/ $\mu\text{L}$ ) oligonucleotide control primer #2
1	dNTP mix
38.5	ddH <sub>2</sub> O

Table 2.2: Mutant Reaction

Volume ( $\mu\text{L}$ )	Reagents
5	10x reaction buffer
3	(5 ng/ $\mu\text{L}$ ) hAR dsDNA template
1.25	(10 $\mu\text{M}$ ) mutant forward primer
1.25	(10 $\mu\text{M}$ ) mutant reverse primer
1	dNTP mix
37.5	ddH <sub>2</sub> O

Afterwards, 1  $\mu\text{L}$  of 2.5U/ $\mu\text{L}$  *PfuUltra* High-Fidelity DNA polymerase was added to each reaction and PCR was initiated with the following thermal cycling parameters (Table 2.3):

Table 2.3: Thermal Cycling Parameters

Cycles	Temperature ( $^{\circ}\text{C}$ )	Time
1	95	30 s
25	95	30 s
	55	1 min
	68	7 min
1	4	30 min

After thermal cycling, the mutant and control reactions were treated with 1  $\mu\text{L}$  of 10 U/ $\mu\text{L}$  of *Dpn* I restriction enzyme to digest the unmutated, methylated DNA. Next, the reactions were gently mixed by pipetting up and down and incubated in the thermal cycler at 37 $^{\circ}\text{C}$  for 1 hour. All *Dpn* I-treated DNA was stored at -20 $^{\circ}\text{C}$ .

## VII. Preparation of -80 $^{\circ}\text{C}$ Bacterial Glycerol Stock

With a sterile wire loop, splinters of ice containing the cells were streaked onto a LB-Amp plate. The plate was incubated overnight at 37 $^{\circ}\text{C}$ . A colony was selected from the plate and it was inoculated in 5 mL of LB-Amp. The cells were incubated at 37 $^{\circ}\text{C}$  and 200 rpm until the Abs<sub>600</sub> reached 0.4-0.6. 4 mL of cell culture was added to 4 mL of 40% v/v glycerol and 800  $\mu\text{L}$  of the bacterial glycerol stocks were dispensed into 10 microcentrifuge tubes. These tubes of bacterial glycerol stocks were stored at -80 $^{\circ}\text{C}$ .

### **VIII. Preparation of Competent Cells with Calcium Chloride Treatment**

This following protocol was used to make the XL1-Blue and BL21(DE3) cells transiently permeable to DNA. With a sterile wire loop, splinters of ice from the bacterial glycerol stock were inoculated in 100 mL LB-Amp (in 500 mL Erlenmeyer flask). The cells were incubated at 37°C and 200 rpm until the Abs<sub>600</sub> was 0.4-0.6. The cell culture was chilled on ice for 30 minutes and then transferred into two 50 mL graduated centrifuge tubes. The cell culture was centrifuged at 4°C and 5,000 rpm for 10 minutes and the supernatant was discarded. The cell pellet was resuspended in 35 mL of 0°C 100 mM calcium chloride solution, gently tapped and incubated on ice for 20 minutes. This mixture was centrifuged at 4°C and 5,000 rpm for 10 minutes and the pellet was resuspended in 1.5 mL of cold (0°C) 100 mM calcium chloride solution. For long-term storage, this cell culture mixture was kept at -80°C in 40% v/v glycerol that was 1:1 with the cell culture.

### **IX. XL1-Blue *E. coli* Transformation**

The XL1-Blue glycerol stock was gently thawed on ice and 100 µL was aliquoted into two prechilled microcentrifuge tubes: one for the mutagenesis control reaction and one for the mutant reaction. Also, 50 µL of XL1-Blue supercompetent cells was aliquoted into one prechilled microcentrifuge tube for the transformation control reaction. The 5-10 µL of *Dpn* I-treated mutant DNA was added to the mutant reaction tube and 5 µL of *Dpn* I-treated control DNA was added to the mutagenesis control tube. For successful transformations, the C298A and C298G mutant reactions required 5 µL of *Dpn* I-treated DNA, while the C298D mutant reaction needed 10 µL of *Dpn* I-treated DNA. No DNA was added to the transformation control tube, but 1 µL of pUC18 control plasmid was

added to it. All three tubes were swirled gently and incubated on ice for 30 minutes. The tubes were heat pulsed for 90 seconds in a 42°C water bath and then placed on ice for 2 minutes. The 450 µL of 42°C preheated SOC media (2% w/v bacto tryptone, 0.5% w/v yeast extract, 0.01 M NaCl, 0.004 mM KCl, 0.01 M MgCl<sub>2</sub>, 0.01 M MgSO<sub>4</sub>, 0.02 M glucose) was added to each tube and they were incubated at 37°C for 1 hour with shaking at 200 rpm. About 250-400 µL of the mutant reaction, 200 µL of the mutagenesis control reaction, and 50 µL of the transformation control reaction were spread onto LB-Amp plates containing X-Gal and IPTG. The plates were incubated overnight at 37°C and successful XL1-Blue transformants appeared white since their *lacZ* gene was disrupted by the insertion of mutant DNA.

#### **X. Isolation of Plasmid DNA (QIAprep ® Spin Miniprep Kit)<sup>14</sup>**

Six white colonies from the XL1-Blue LB-Amp plate were streaked onto one LB-Amp plate (subdivided into 6 sections). Three colonies (one from 3 separate sections) were selected and inoculated in 5 mL of LB-Amp. The cell cultures were incubated at 37°C with 200 rpm shaking overnight. The 4 mL of cell culture from each section was equally dispensed into 3 microcentrifuge tubes. There are now 3 microcentrifuge tubes from each section (9 tubes total) and each tube contains 1.333 mL of cell culture. The tubes were centrifuged at 8,000 rpm and 25°C for 3 minutes. The pellets from each section were combined, resuspended in 250 µL of Buffer P1 (resuspension buffer) and transferred to a microcentrifuge tube. The following procedures refer to what was performed to each of the 3 microcentrifuge tubes. For the lysis reaction, 250 µL of Buffer P2 (lysis buffer) was added and the mixture was gently mixed until it turned clear. For the neutralization reaction, 350 µL of Buffer N3 (neutralization buffer) was added and

the tube was gently inverted 5 times. Later, the solution was centrifuged at 13,000 rpm and 25°C for 10 minutes. All subsequent centrifugations were done at 13,000 rpm and 25°C. The supernatant was pipetted into a QIAprep spin column, centrifuged for 1 minute and the flow-through was discarded. The 500 µL of Buffer PB (binding buffer) was added to the spin column, it was centrifuged for 1 minute and the flow-through was discarded. The 750 µL of Buffer PE (wash buffer) was added to the spin column, it was centrifuged for 1 minute and the flow-through was discarded. To remove residual buffer, the spin column was centrifuged for 1 minute and the flow-through was discarded. The spin column was placed in an empty microcentrifuge tube and 50 µL of Buffer EB (10 mM Tris-Cl, pH 8.5) was added to elute the DNA. The tube was left undisturbed for 1 min and centrifuged for 1 min. The isolated DNA was stored at -20°C.

### **XI. DNA Sequencing**

Nine tubes of isolated DNA (3 from each Cys298 mutant) were sent to Eurofins Genomics for DNA sequencing. After DNA sequencing, the sequences of mutants were analyzed by BioEdit Sequence Alignment Editor Version 7.2.5 to check for the correct mutation and quality of the sequences. With this software, a ClustalW multiple sequence alignment was performed to compare each mutation sequence with the hAR wild-type sequence. The sequences that matched more closely with the wild-type were chosen and used to select the *Dpn* I-treated DNA samples for the transformation of BL21(DE3) competent cells.

## **XII. BL21(DE3) *E. coli* Transformation**

The 2  $\mu\text{L}$  of Dpn I-treated DNA was added to 50-200  $\mu\text{L}$  of BL21(DE3) glycerol stock (in microcentrifuge tube) and incubated on ice for 30 minutes. The C298A and C298D mutant reactions required 50  $\mu\text{L}$  of bacterial glycerol stock, while the C298G mutant reaction required 200  $\mu\text{L}$  of bacterial glycerol stock. Next, the mutant reaction was heat pulsed for 90 seconds at 42°C and then placed on ice for 2 minutes. The 450  $\mu\text{L}$  of preheated 42°C SOC media was added to the mutant reaction and it was incubated at 37°C for one hour. 50-75  $\mu\text{L}$  of the reaction was spread onto a LB-Amp plate. The plates were incubated overnight at 37°C.

## **XIII. SDS-PAGE**

Protein samples were resolved on a 10% w/v sodium dodecyl sulfate (SDS) polyacrylamide gel. Below are the protocols for preparing the separating and stacking gel (Table 2.4 and Table 2.5).



Table 2.4: Separating Gel (10 mL)

Reagents	Volume (mL)
ddH <sub>2</sub> O	3.8
acrylamide/bisacrylamide liquid (37.5:1)	3.4
Resolving Gel Buffer (1.5 M Tris-HCl, pH 8.8)	2.6
10% w/v SDS	0.1
10% w/v ammonium persulfate	0.1
TEMED	0.01

Table 2.5: Stacking Gel (5 mL)

Reagents	Volume (mL)
ddH <sub>2</sub> O	2.975
acrylamide/bisacrylamide liquid (37.5:1)	0.67
Stacking Gel Buffer (0.5 M Tris-HCl pH 6.8)	1.25
10% w/v SDS	0.05
10% w/v ammonium persulfate	0.05
TEMED	0.005

The running buffer was prepared by diluting the 10x Tris/Glycine/SDS Electrophoresis Buffer (25 mM Tris, 192 mM glycine, 0.1% w/v SDS, pH 8.3) to 1x with ddH<sub>2</sub>O. The staining solution was prepared with 0.3% w/v Coomassie Brilliant Blue G-250, 50% v/v

methanol, 40% v/v H<sub>2</sub>O, and 10% v/v acetic acid. The de-staining solution was prepared with 30% v/v methanol, 60% v/v H<sub>2</sub>O, and 10% v/v acetic acid. The 4x SDS protein sample buffer was prepared with 40% v/v glycerol, 240 mM Tris (pH 6.8), 8% w/v SDS, 0.04% w/v bromophenol blue, and 5% v/v BME. The 2x SDS protein sample buffer was prepared by diluting the 4x SDS protein sample buffer with 1x PBS buffer. The 1x phosphate buffered saline (PBS) buffer was prepared with 0.8% w/v sodium chloride, 0.02% w/v potassium chloride, sodium phosphate 0.144% w/v, and 0.024% w/v potassium diphosphate with the pH adjusted to 7.4. Protein gel samples from protein purification were prepared by mixing 75  $\mu$ L of the sample with 25  $\mu$ L of the 4x SDS protein sample buffer. Protein gel samples from protein expression were prepared by adding 60  $\mu$ L of the 2x SDS protein sample buffer to the cell pellets (1 mL of cell culture that had been centrifuged for 2 minutes). PageRuler Plus Prestained Protein Ladder was used as the molecular weight marker and all gel samples were boiled for 3 min before being loaded into the wells. Initially, the electrical voltage was set to 70 V, but once the separating gel had been crossed it was increased to 100 V. Once the protein samples had run the entire length of the gel, the gel was removed and rinsed with water. The gel was placed in the staining solution and microwaved for 45 seconds on high power. It was placed on a low speed laboratory shaker to shake for 5 minutes. Next, the gel was placed in the de-staining solution, microwaved for 45 seconds on high power and gently shaken for 5 minutes. This cycle was repeated 3-4 times with fresh de-staining solution and then left to gently shake overnight. The following day the bands on the gel were visualized with an ultraviolet illuminator. For long-term storage, SDS gels were kept submerged in water.

## **Chapter III: Protein Expression**

### **I. Protein Expression**

A single colony of BL21(DE3) was picked from a LB-Amp plate and inoculated in 50 mL of 2YT media (in 125 mL Erlenmeyer flask). It was incubated overnight at 37°C and 200 rpm. The 10 mL of the cell culture was inoculated in 1 L of 2YT media (in 2,800 mL Erlenmeyer flask) and incubated for 2.5-3 hours at 37°C and 200 rpm. Once the Abs<sub>600</sub> reached 0.6-0.8, protein expression was induced with 10 mL of 100 mM IPTG and the cell culture was incubated for 4 hours at 37°C and 200 rpm. Gradually, the cell culture was added to two 250 mL centrifuge bottles and centrifuged at 4°C and 6000 rpm in 5 minute cycles. The collected cell pellets were stored at -80°C.

## **Chapter IV: Protein Purification**

### **I. Sonication**

The cell pellets were thawed on ice and resuspended with the extraction buffer to yield a total volume of 60-80 mL. This suspension was mixed with a vortex mixer and pipetted until the solution had no clumps. The suspension was poured into a 100 mL beaker and this beaker was then placed inside a 1 L beaker that was filled with ice. The 1 L beaker was placed inside the sonicator and the probe was lowered into the smaller beaker until it was within a ½ inch from the bottom of the beaker. The suspension was sonicated in a 30 second burst and allowed to cool for 30 seconds. This first cycle was repeated 15 times. Next, the suspension was sonicated in a 1 minute burst and allowed to cool for 1 minute. This second cycle was repeated 10 times. The sonicated suspension was centrifuged at 12,000 rpm and 4°C for 30 minutes. The supernatant was collected in a plastic bottle and the pellets were combined and resuspended with extraction buffer. For

the resuspended pellets, the entire sonication and centrifugation process was repeated as before and the all the supernatant was collected in one plastic bottle.

## **II. Immobilized Metal Affinity Chromatography**

The 5 mL of Talon CellThru Immobilized Metal Affinity Chromatography (IMAC) resin was washed with 15 mL of extraction buffer. This process was repeated 3 times and each time the liquid was discarded with a syringe. This resin was added to the supernatant and it was gently shaken with a small laboratory rocker for 12-16 hours at 4°C. This procedure was done to bind the affinity resin with the histidine tag of the protein. A 30 mL gravity-flow column was washed with these solutions in the following order: 50 mL of 2 M NaOH, 50 mL of ddH<sub>2</sub>O, and 25 mL of extraction buffer. In a 4°C cold room, the supernatant/resin mixture was added to the column and the flow-through was collected at a slow, steady rate. The 50 mL wash 1 buffer was added to the column and wash 1 was collected in one fraction. The 50 mL of wash 2 buffer was added to the column and wash 2 was collected in one fraction. The 50 mL of elution buffer was added to the column and 5 eluates were collected in 10 mL fractions. All of the fractions were stored at 4°C.

## **III. Dialysis**

SDS-PAGE was used to confirm which of the elution fractions had the most protein and these fractions were loaded into Spectra/Por 1 dialysis tubing. The protein was dialyzed for 8-12 hours in 2 L of dialysis buffer that was slowly stirred at 4°C. The dialysis buffer was changed for a total of 3 times and the protein was transferred to 50 mL graduated centrifuge tubes.

#### **IV. Protein Quantification: Bradford Reagent Assay**

The 20% v/v dye reagent was prepared by diluting the Dye Reagent Concentrate with ddH<sub>2</sub>O and filtering it with filter paper. Six dilutions of IgG protein standard were prepared that ranged from 0.2 mg/mL to 1.2 mg/mL. The 100  $\mu$ L of protein sample and 100  $\mu$ L of IgG protein standard were pipetted into glass tubes. The 5 mL of 20% v/v dye reagent was added to each tube and they were vortexed and incubated at 37°C for 5 minutes. The absorbance at 595 nm was recorded and the protein concentration of the protein sample was calculated from an IgG standard graph of Abs<sub>595</sub> versus concentration.

#### **V. Thrombin Cleavage of Histidine Tag**

After the protein concentration of the dialysis protein sample was determined, 1  $\mu$ L of thrombin for every 4 mg of protein was added to the dialysis protein sample. The pre-digested protein was kept overnight at room temperature. Next, the IMAC resin was added to the digested protein and it was gently shaken with a small laboratory rocker for 12-16 hours at 4°C. This procedure was done to bind the affinity resin with the cleaved histidine tag of the protein. At room temperature the protein/resin mixture was added to the column and the flow-through was collected at a slow, steady rate.

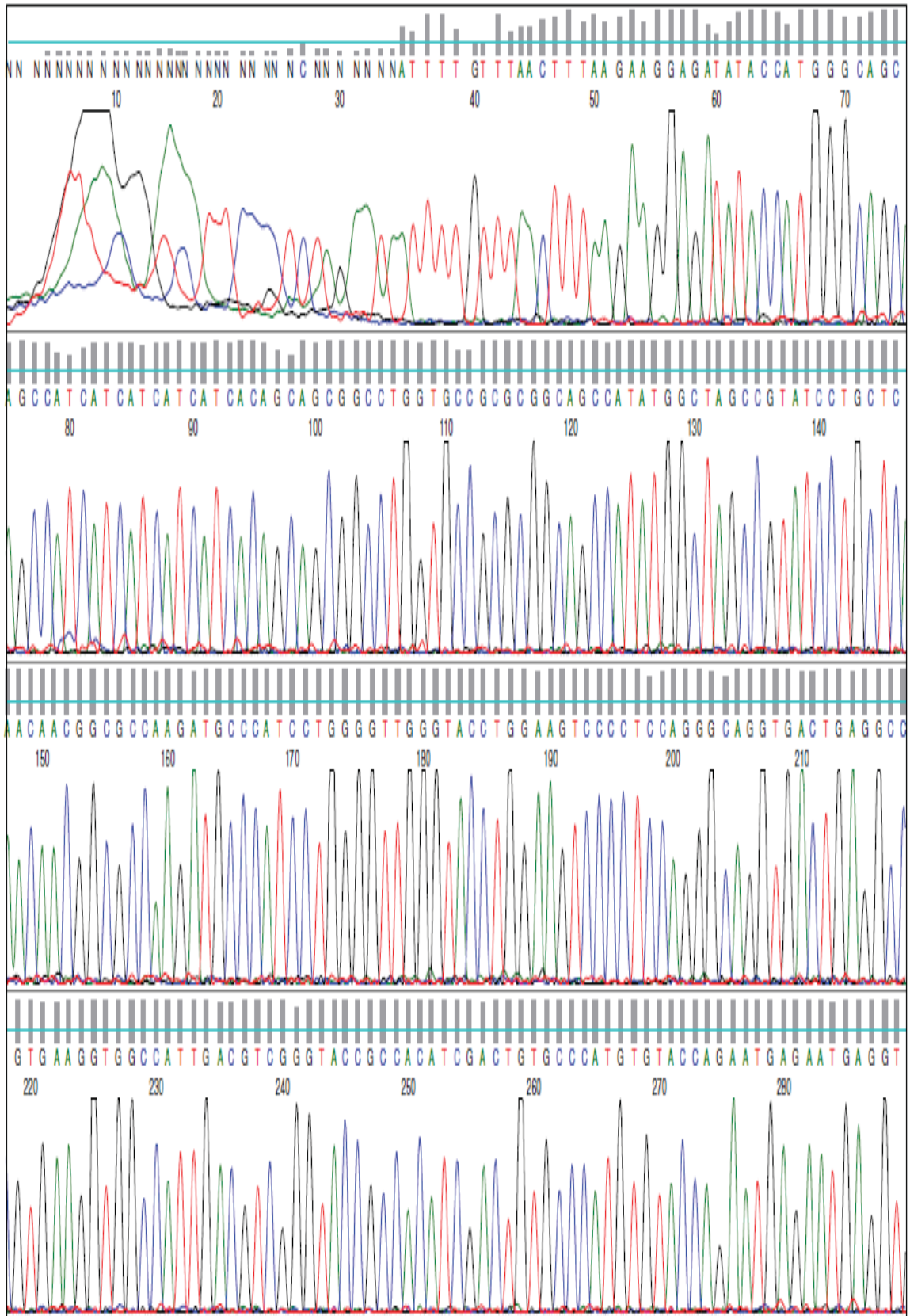
## **Chapter V: Results**

### **I. DNA Sequencing**

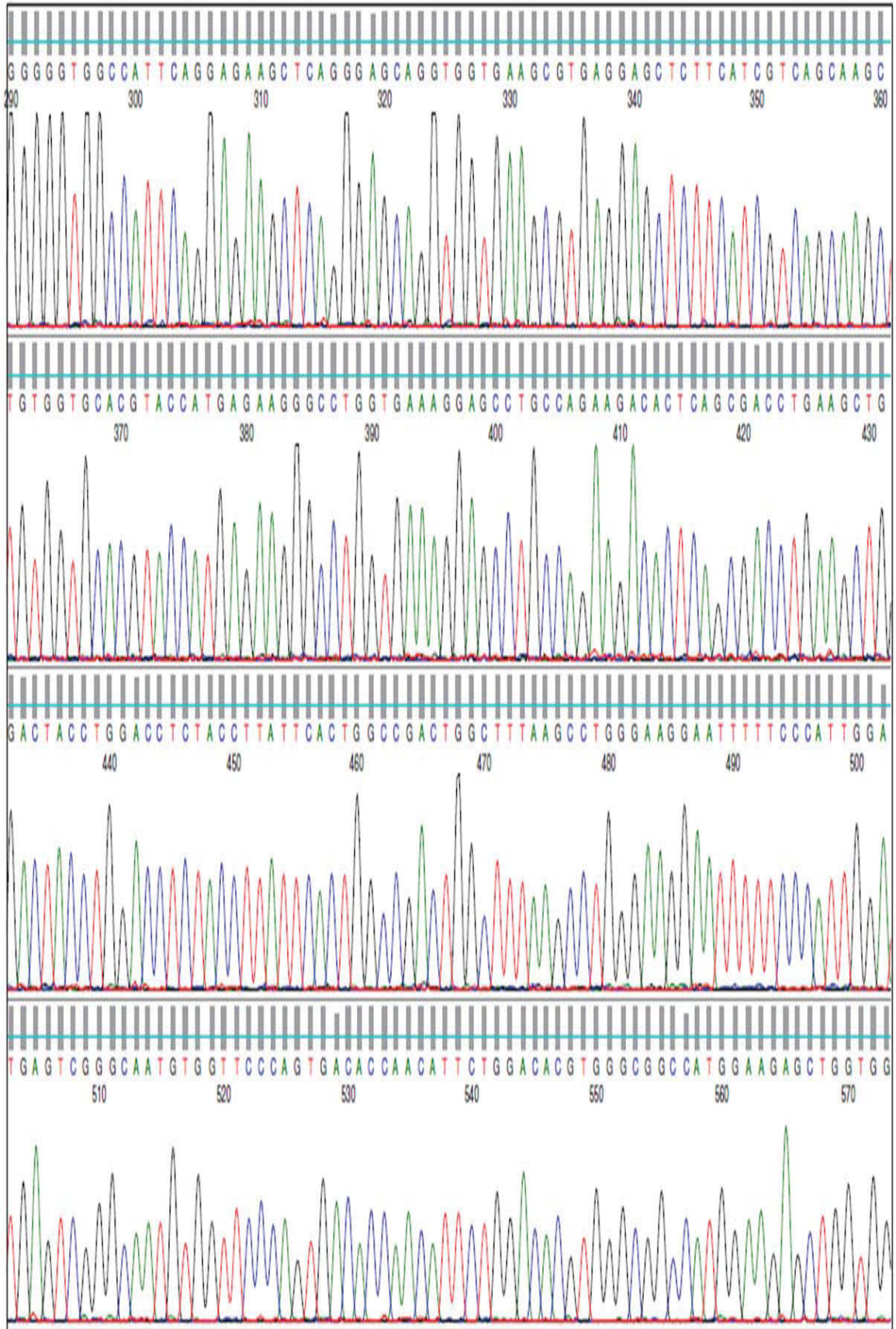
Below, are the DNA sequences (Figures 5.1, 5.3, 5.5, 5.7, 5.9, 5.11) of the isolated DNA that was used for the protein expression of each Cys298 mutant. Underlined in these sequences are the mutation sites. Also, the DNA chromatograms (Figures 5.2, 5.4, 5.6, 5.8, 5.10, 5.12) of each Cys298 mutant have well-resolved peaks and no ambiguities within the protein coding sequences. In the following DNA chromatograms T represents thymidine, A represents adenosine, G represents guanine, C represents cytosine and N represents an undetermined nucleotide.

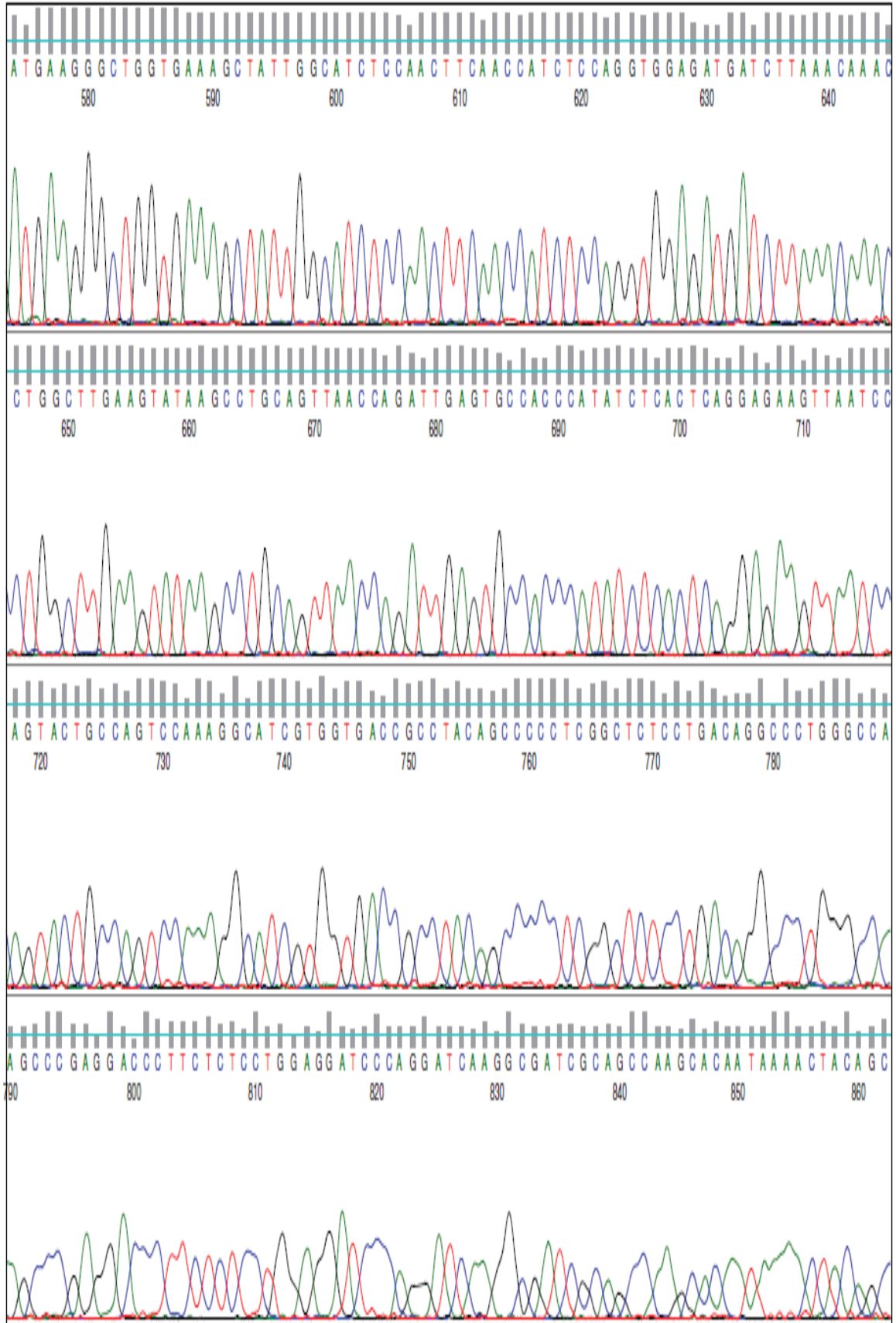


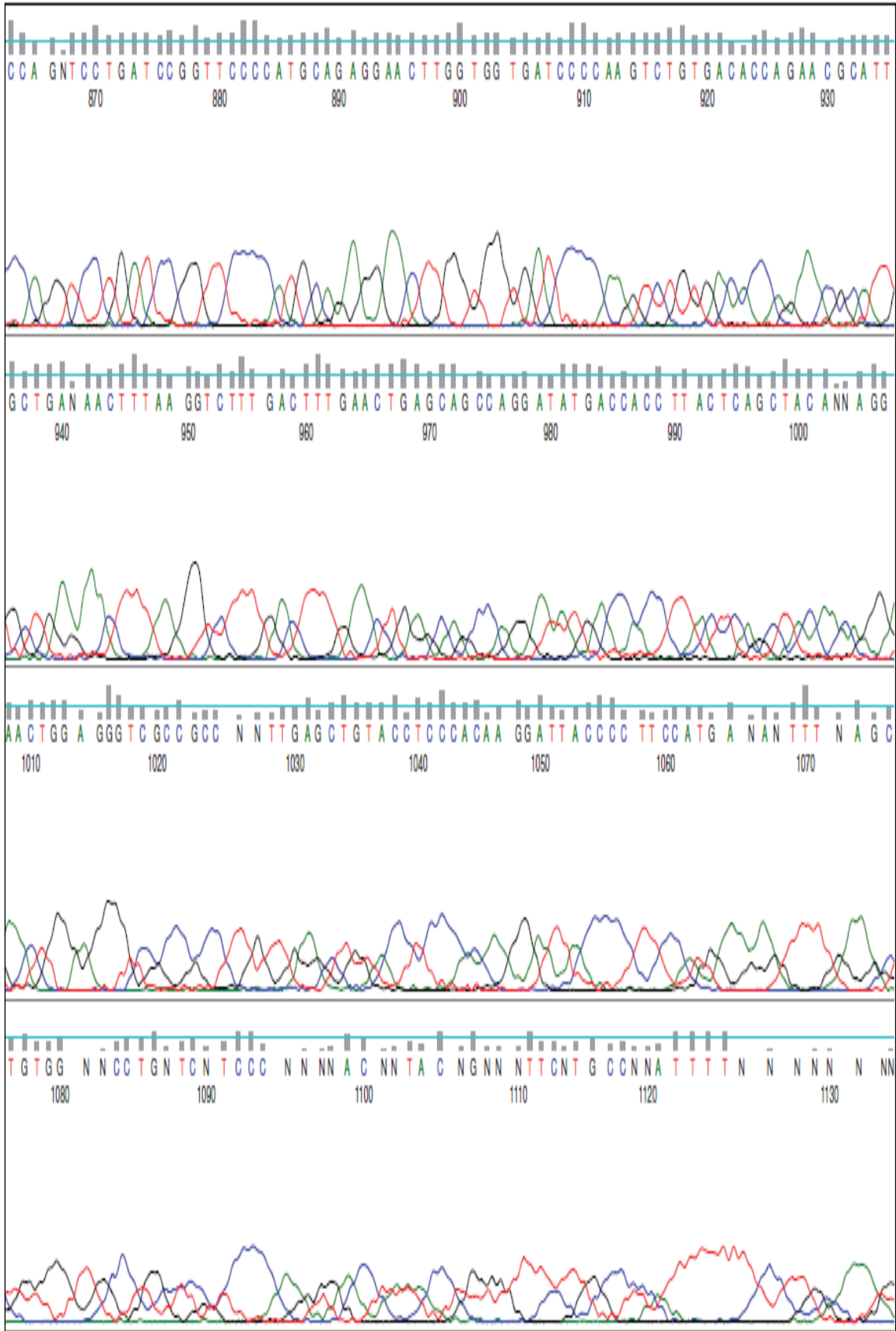
Figure 5.2 C298A Forward DNA Chromatogram











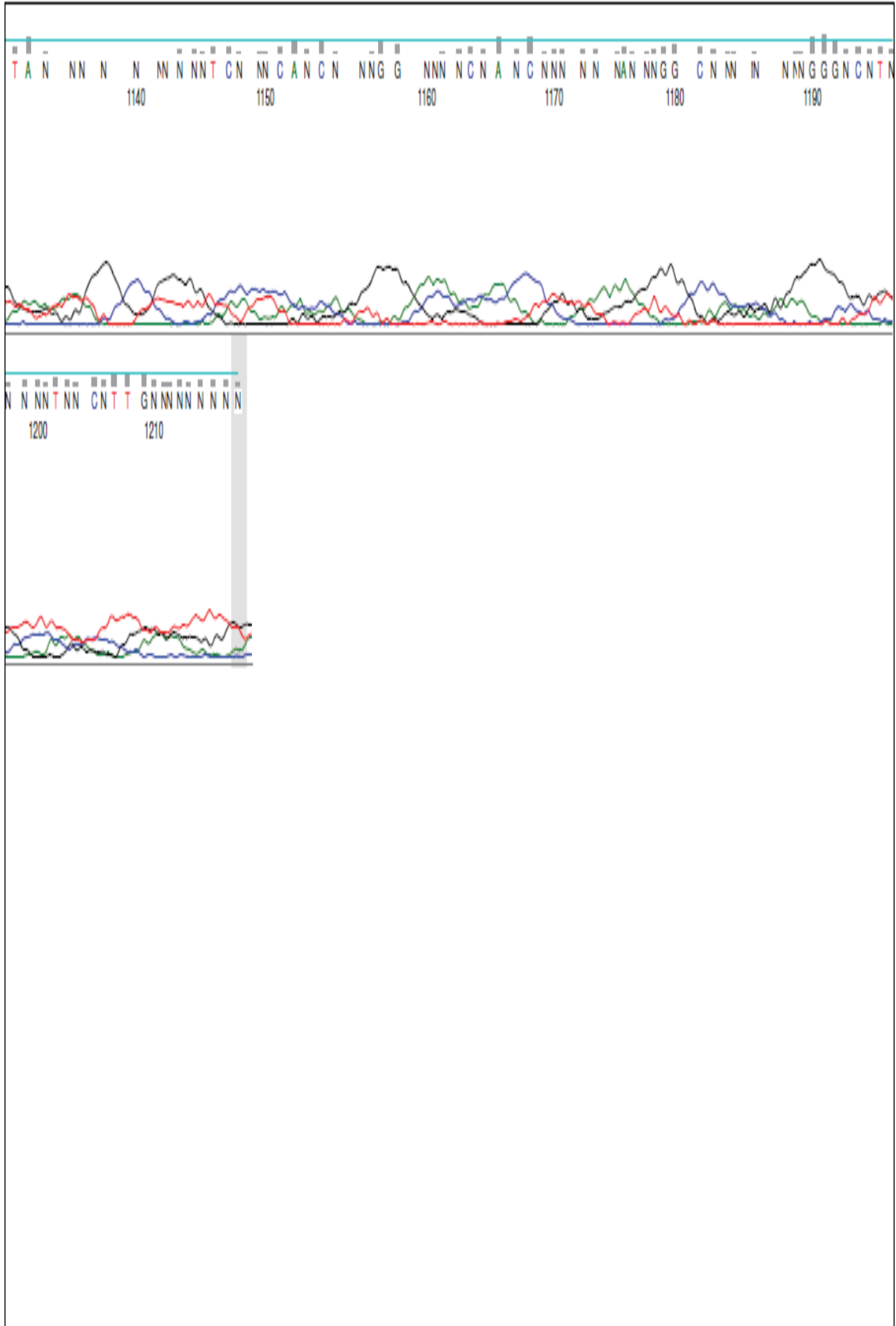
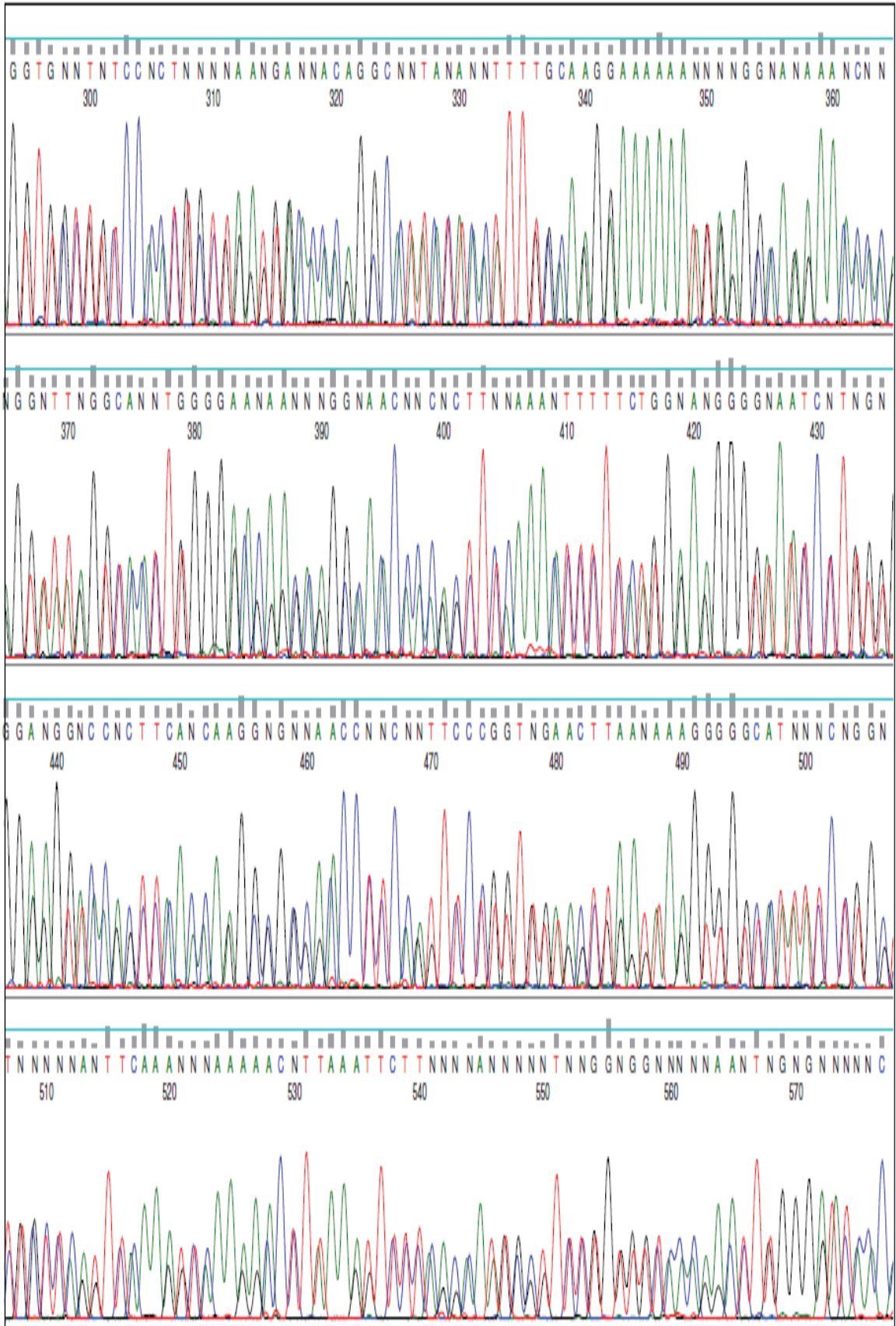
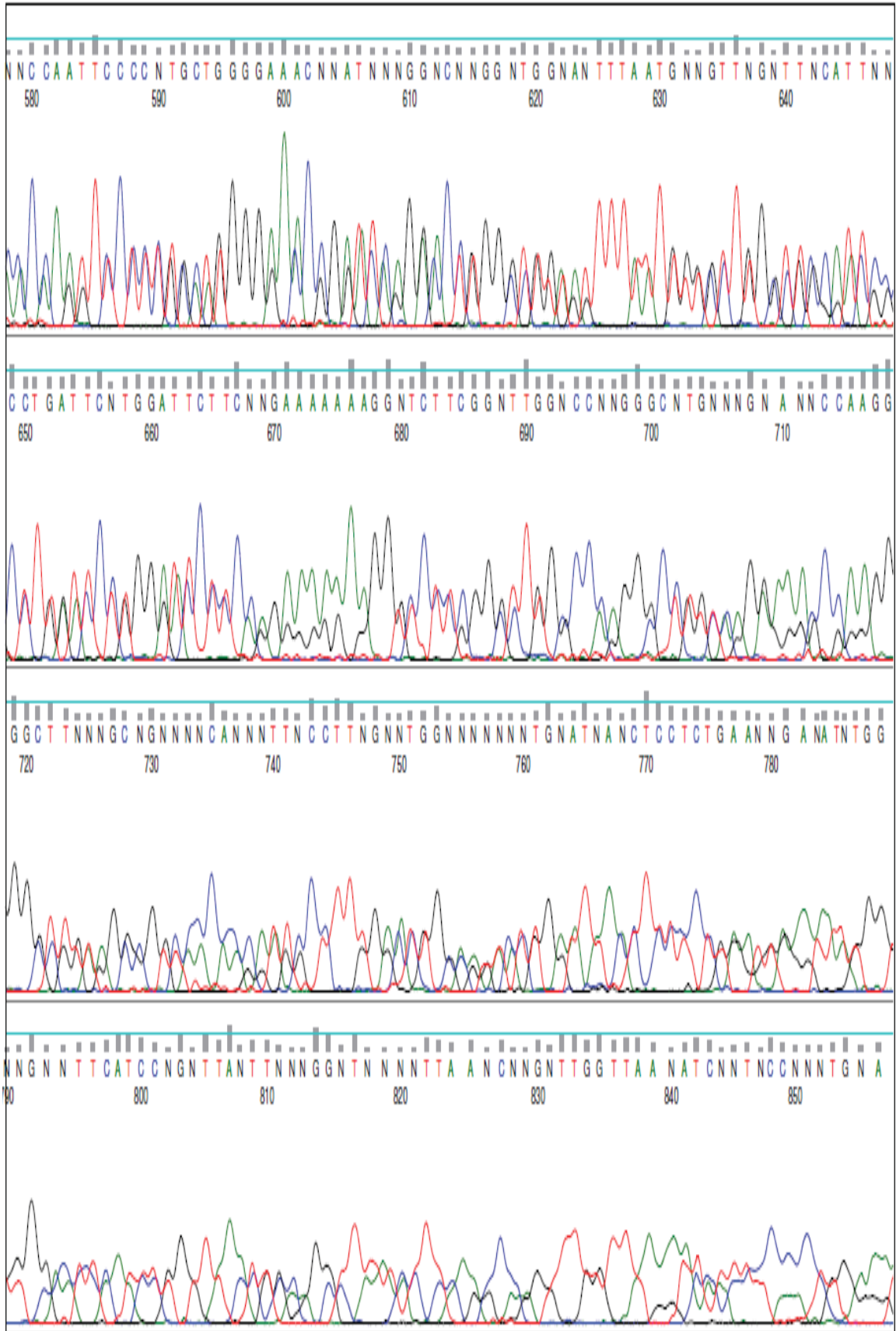


Figure 5.3: C298A Tube 1 Reverse Sequence

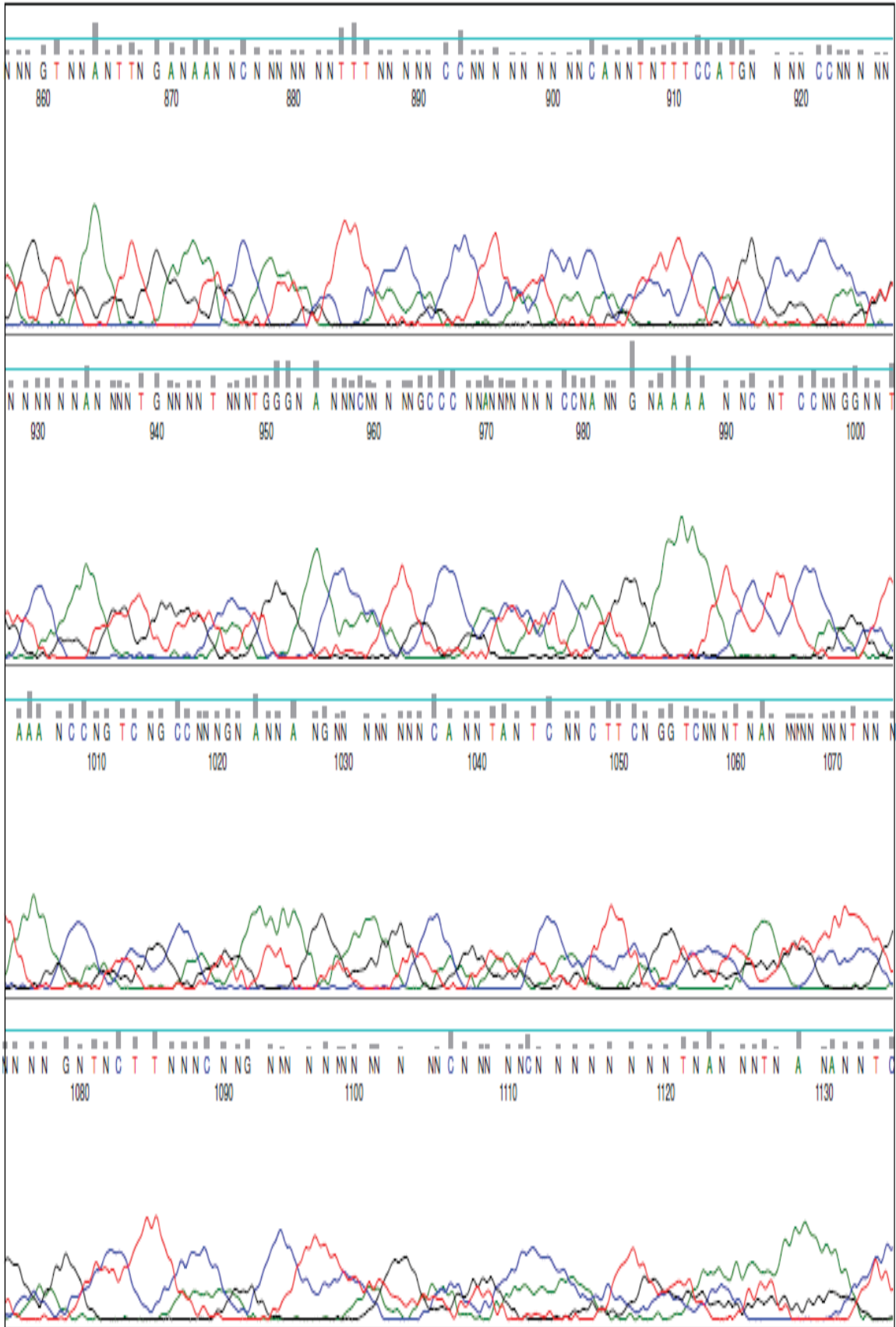
NNNNNNNNNNNNNNNTTTCGGGCTTTGTTAGCAGCCGGATCCTCGAGGCAAAG  
AGAAGTCTTGCTGAAAGGATTCCAGTTCCAAGCAGTCAAACTCAACCGTTA  
GTGGCACTATTTTGACCTGGTAGATTTTGCTTCTCTTTGGTCAGAAAAGGGTA  
TTCAGGTTGTACTTTCCCCAGCAGGGTAGAAAGAAGGGCAAAGCAAAGCTGGA  
AGAGACTTCTACTCTACTGACAGGGCTCTTGAGATCCAACATCAAGCTAGAC  
ACGCCCTCGCTGGCCACTCTACAGGTTGCTGTCCCACTGCTGAGTGACACAG  
GCCATACTACATTTGCAAGGAAAAAAATGAGGCAAGAAACACAGGTATAGG  
TCACTTGGGGACGAGCAGGCAACCACAGCTTCAAACTCTTCATGGAAGGGG  
TAATCCTTGTGGGAGGTACAGCTCAACAAGGCGGCGACCCTCCAGTTCCTGT  
TGTAGCTGAGTAAGGTGGTCATATCCTGGCTGCTCAGTTCAAAGTCAAAGAC  
CTTAAAGTTCTCAGCAATGCGTTCTGGTGTACAGACTTGGGGATCACCACCA  
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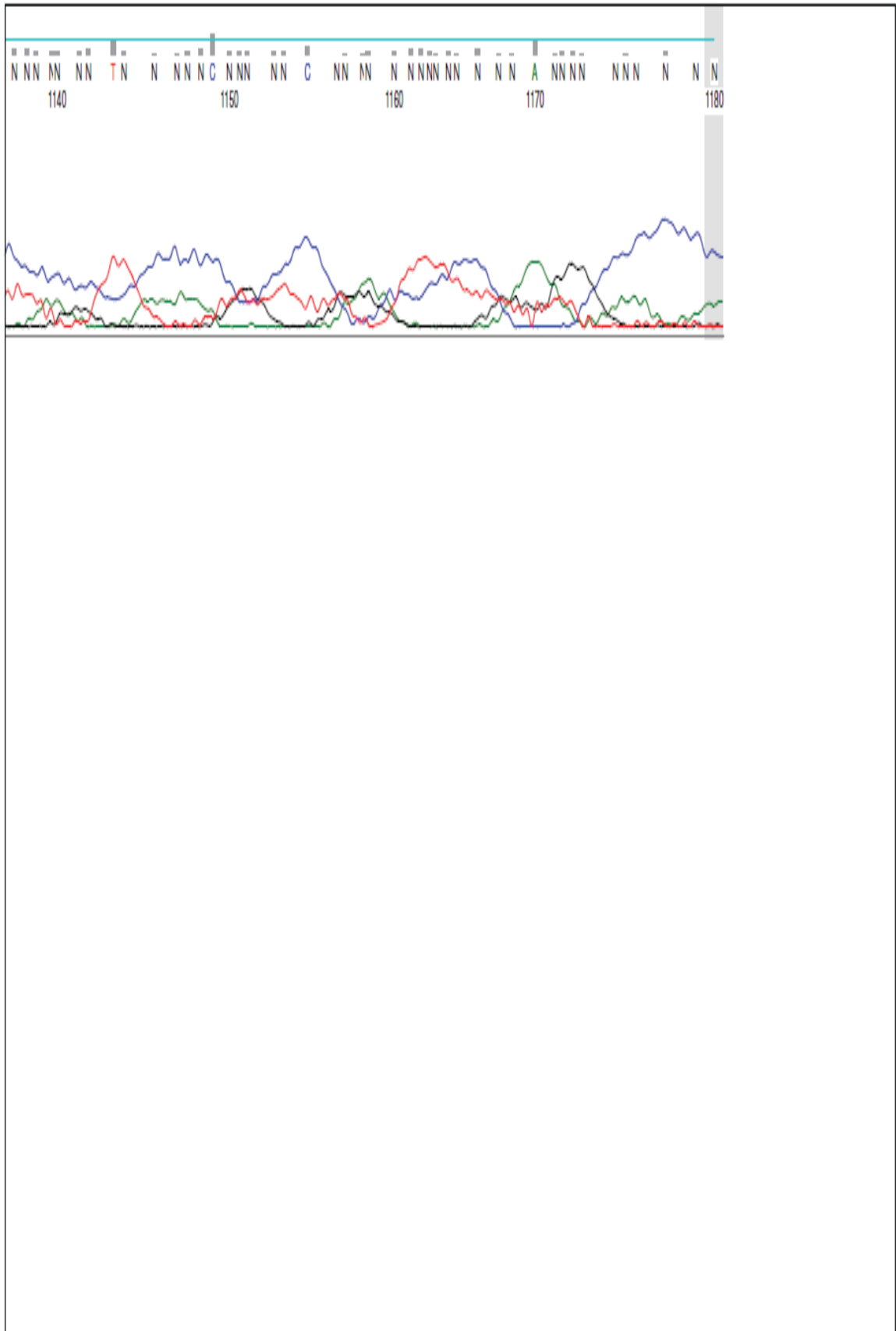
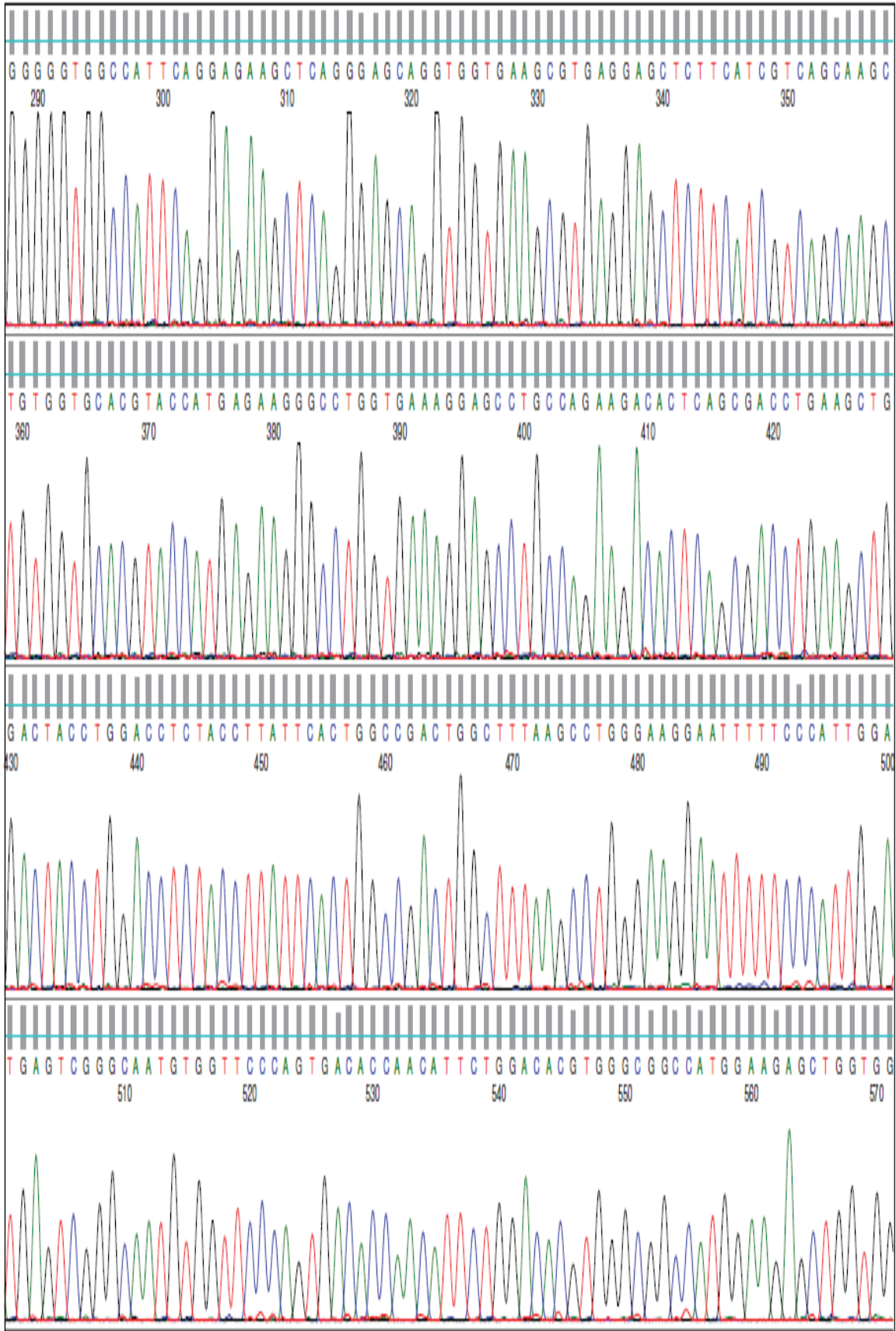
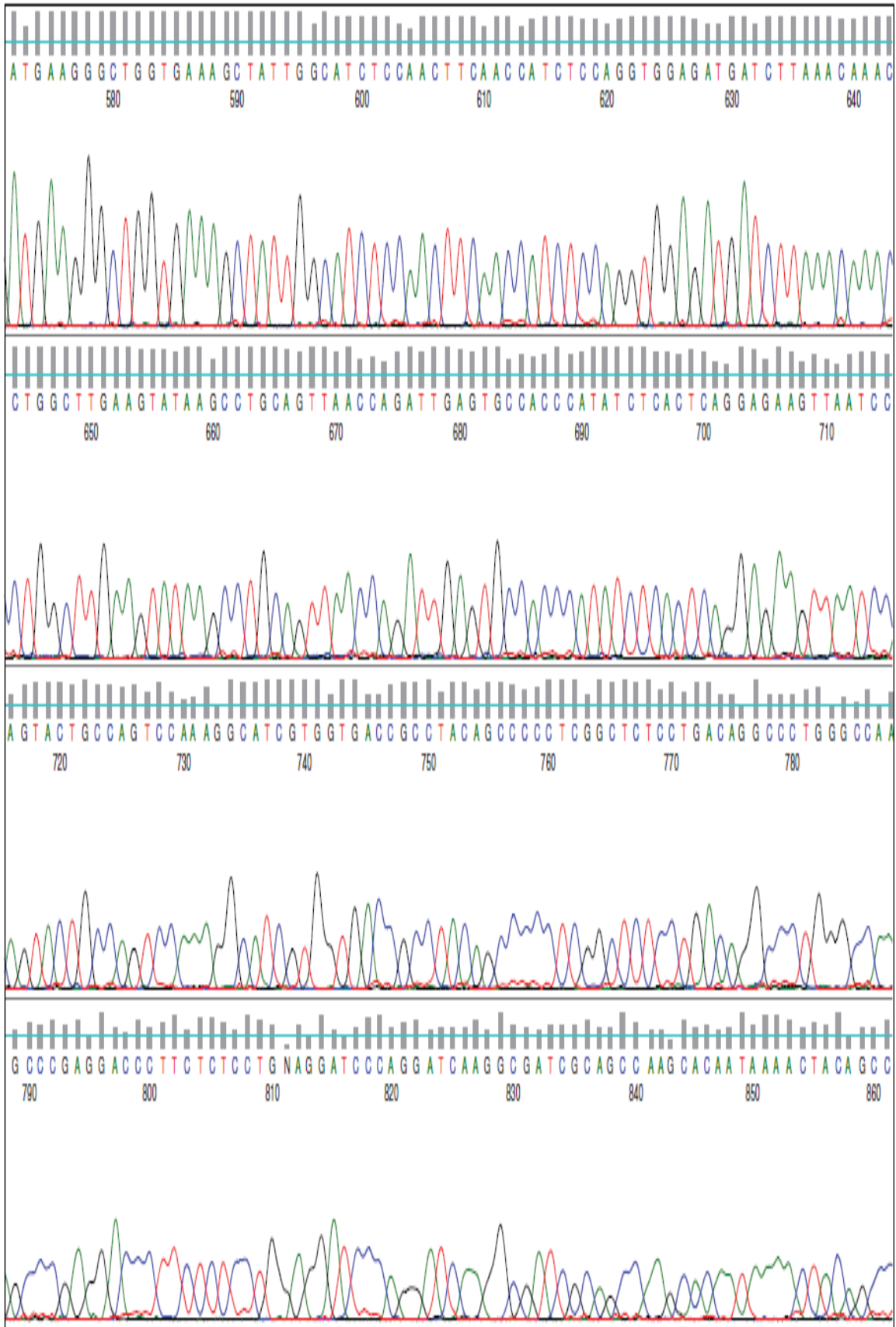


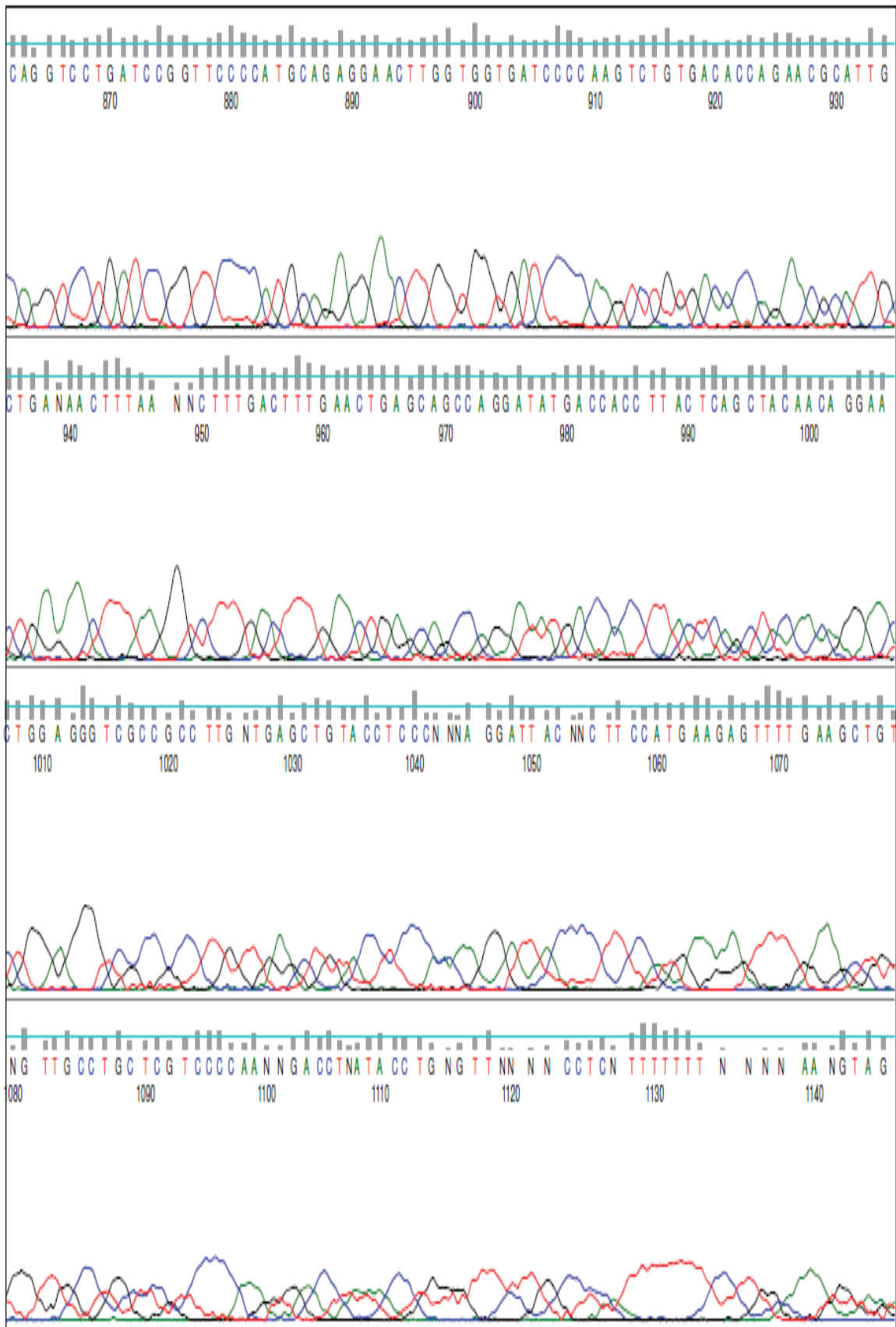
Figure 5.5: C298D Tube 2 Forward Sequence

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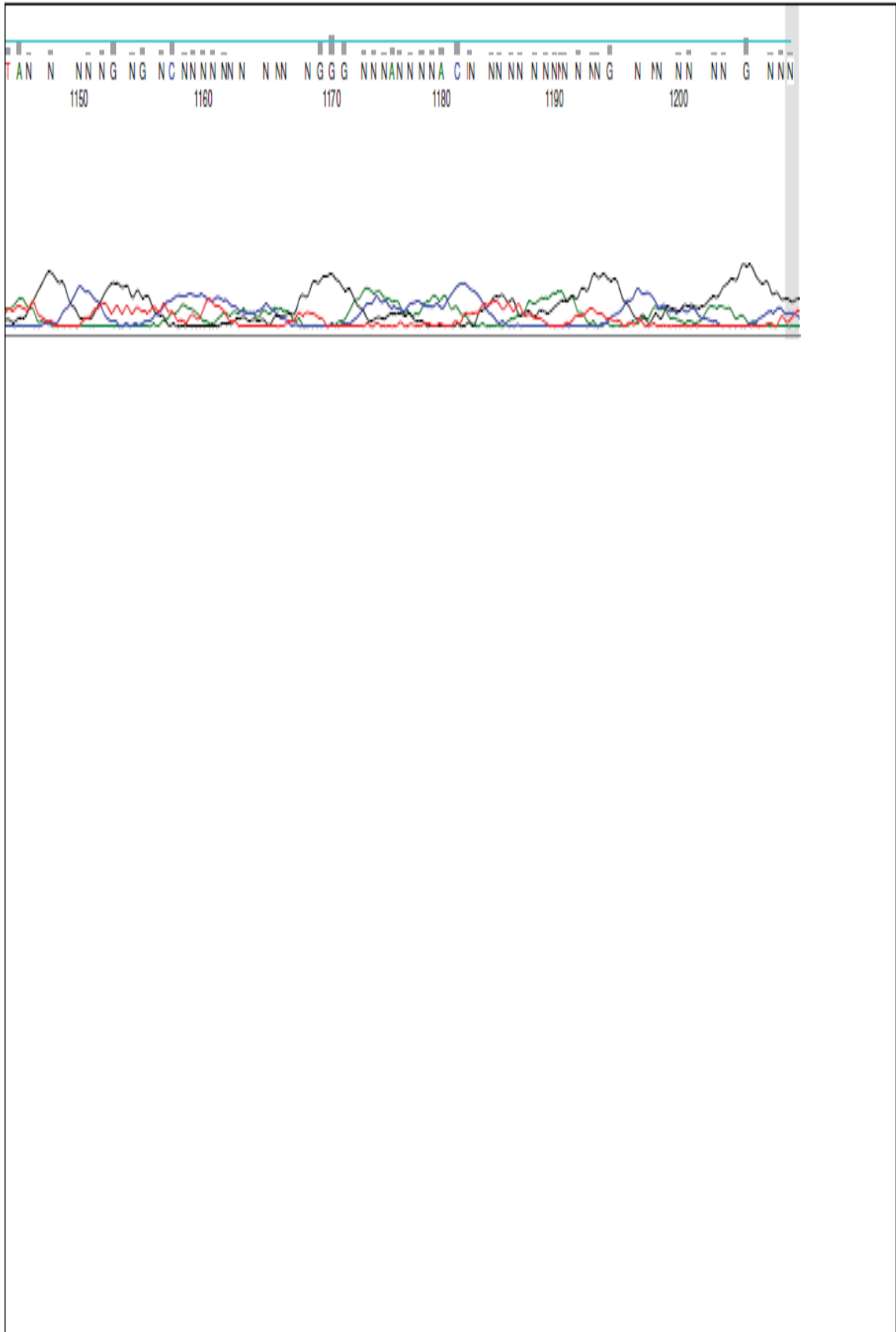
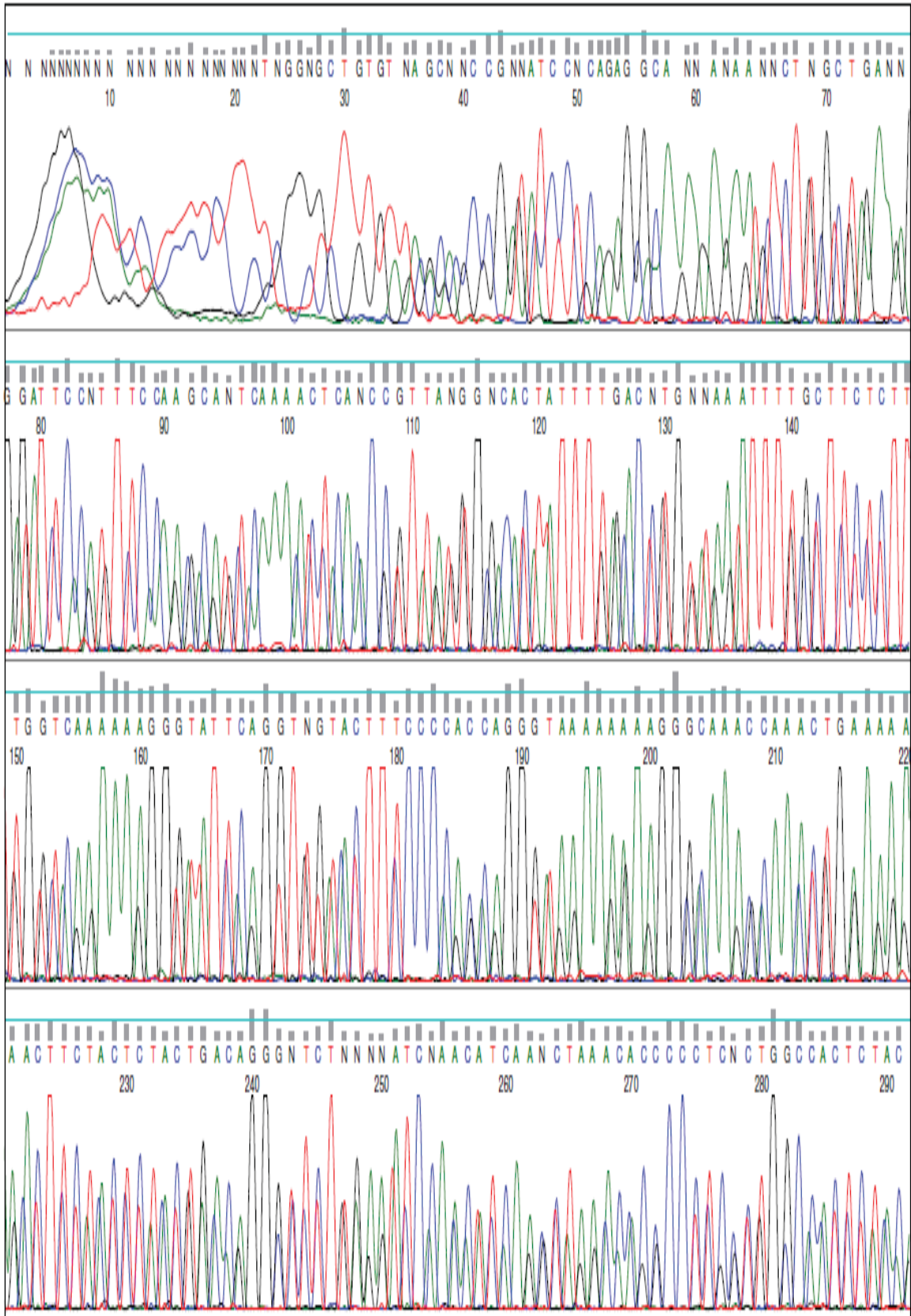


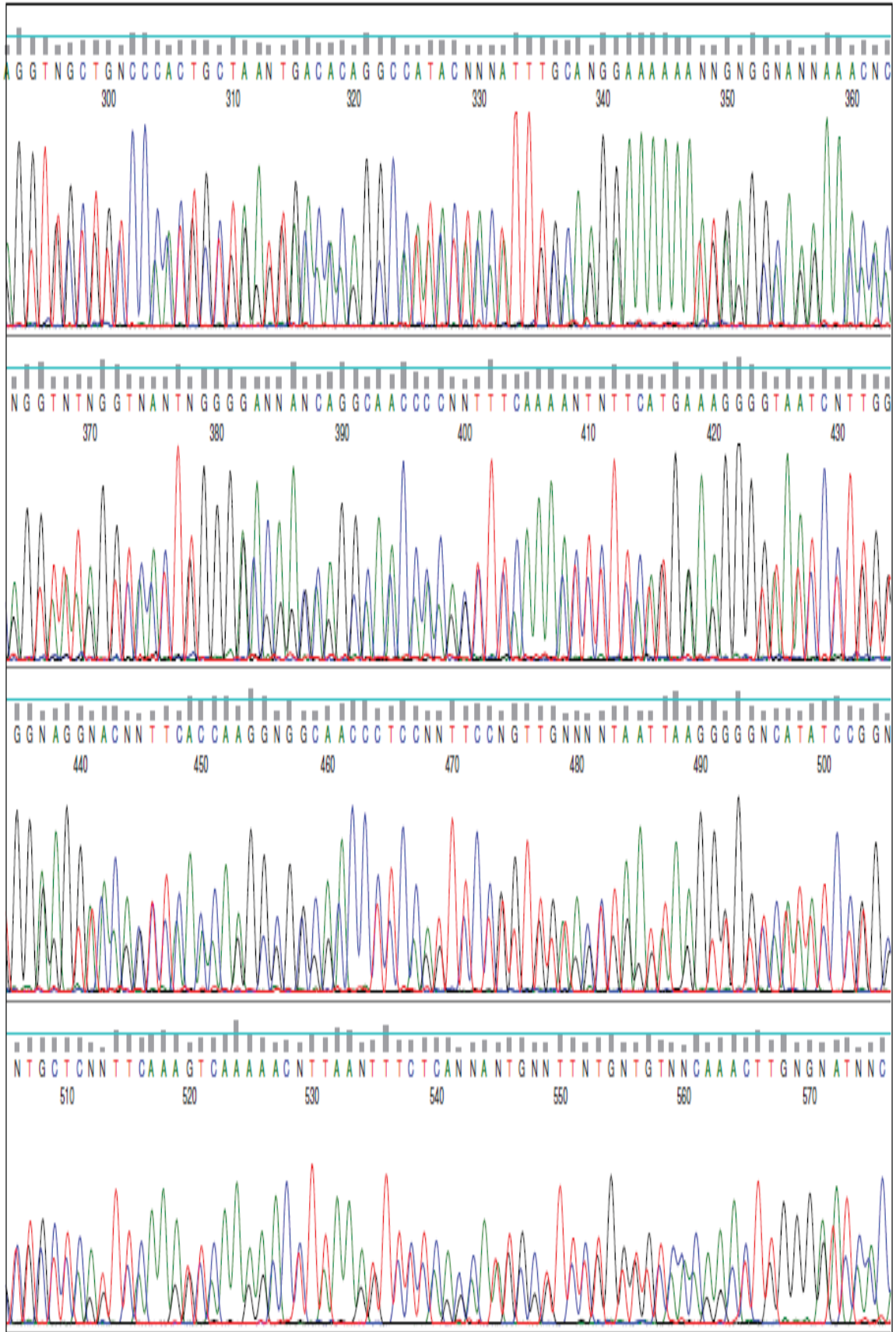


Figure 5.7: C298D Tube 2 Reverse Sequence

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TTCAGGTTGTACTTTCCCCAGCAGGGTAGAAAGAAGGGCAAAGCAAAGTGGAA  
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Figure 5.8: C298D Reverse DNA Chromatogram





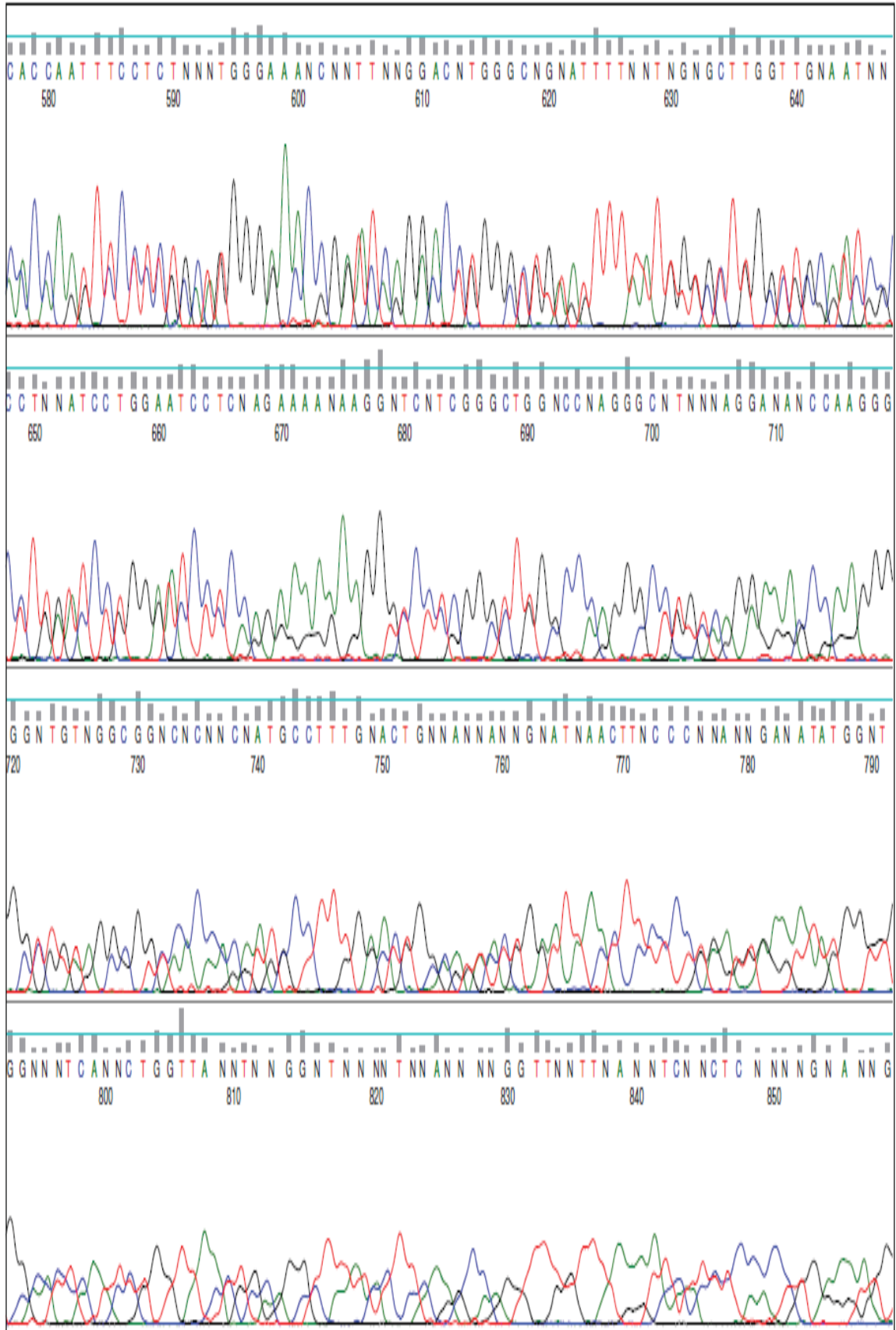


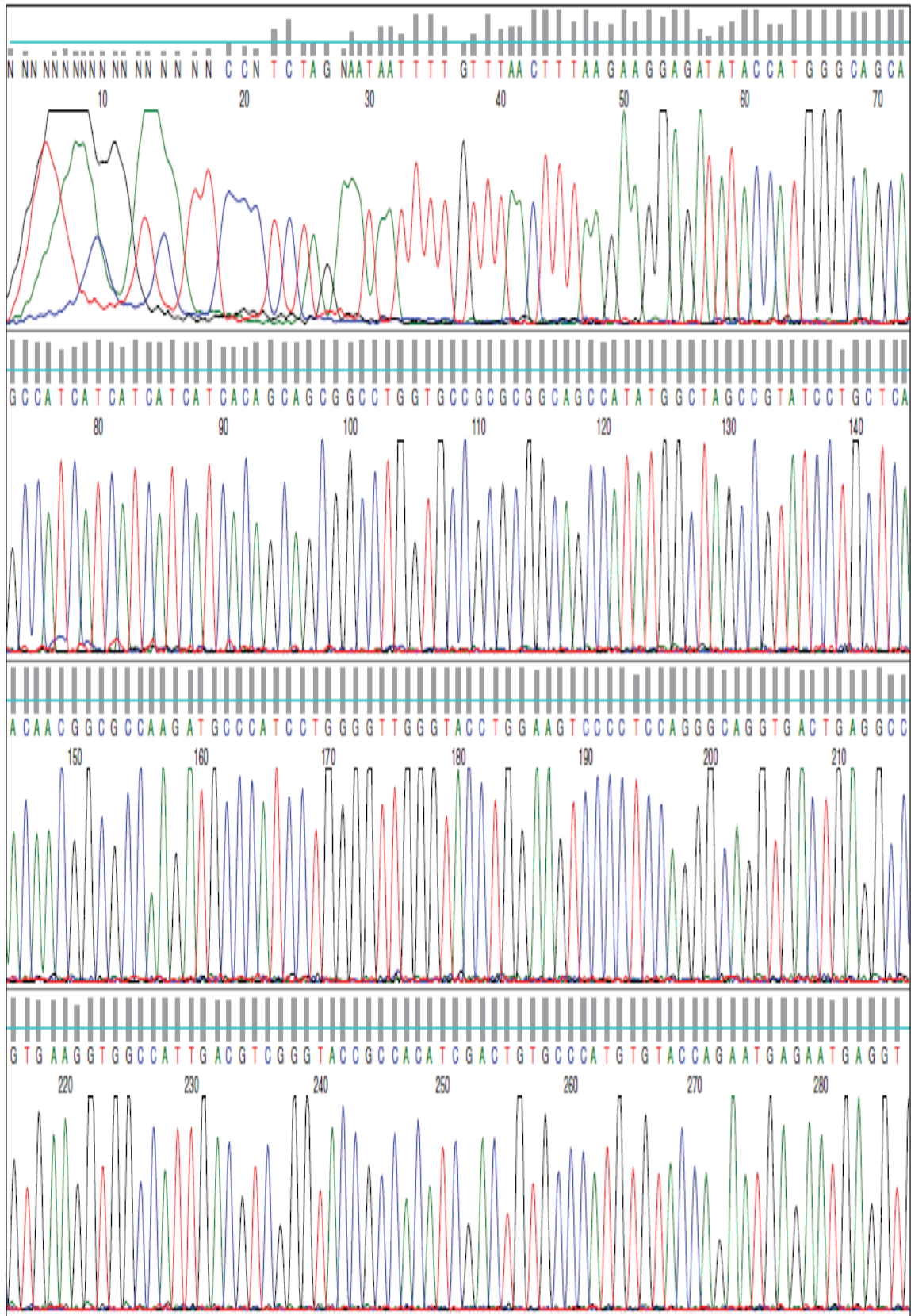




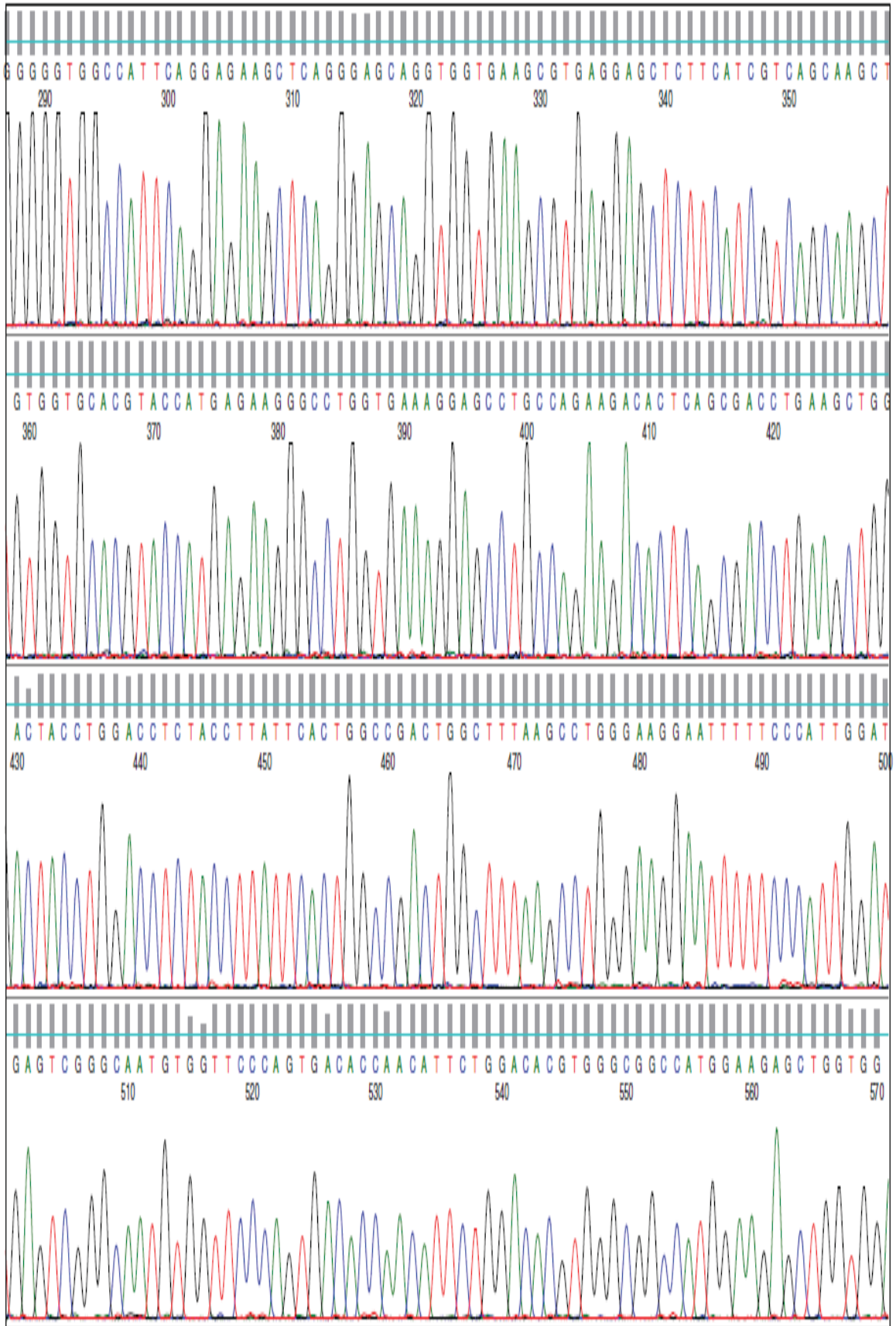
Figure 5.9: C298G Tube 1 Forward Sequence

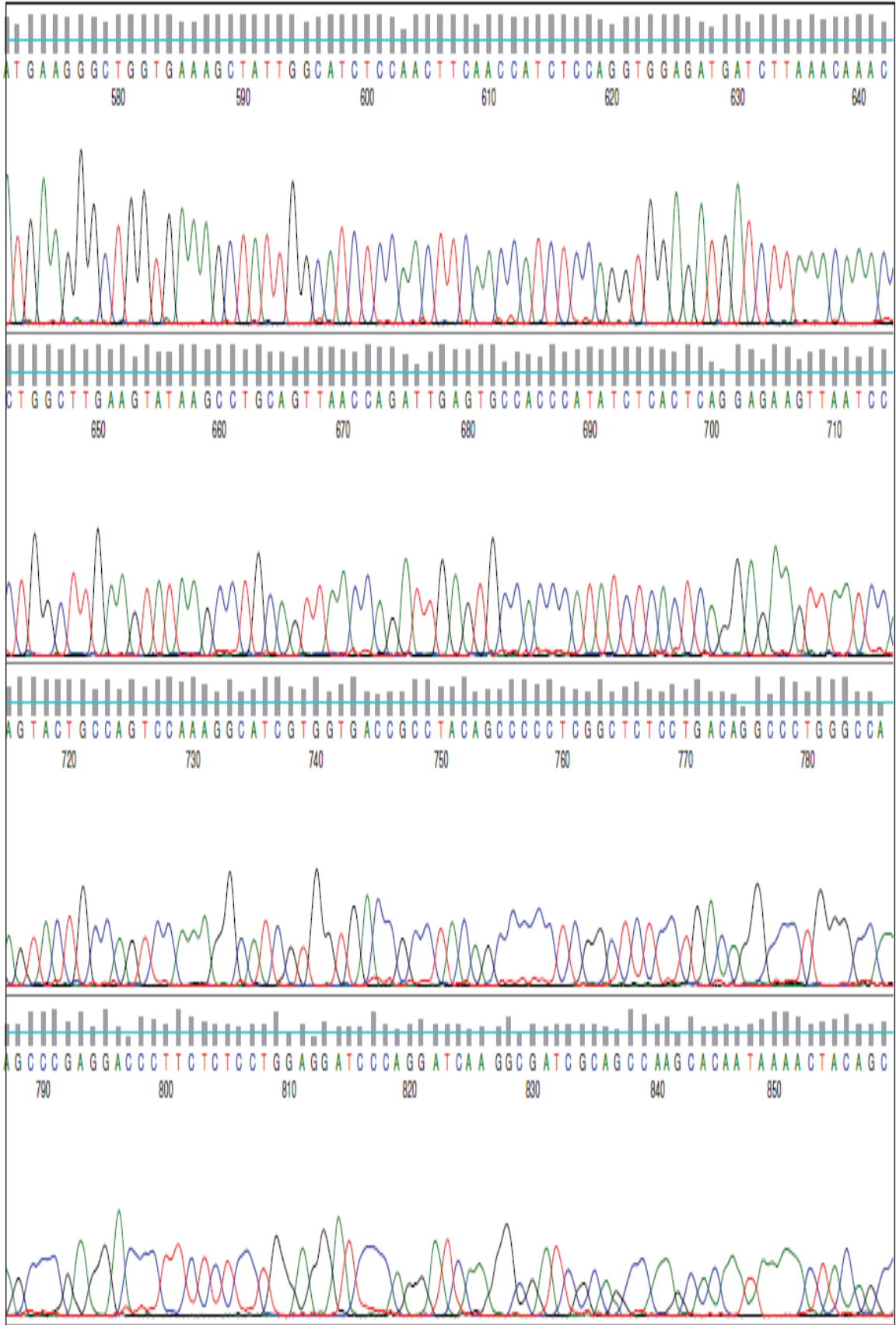
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Figure 5.10: C298G Forward DNA Chromatogram









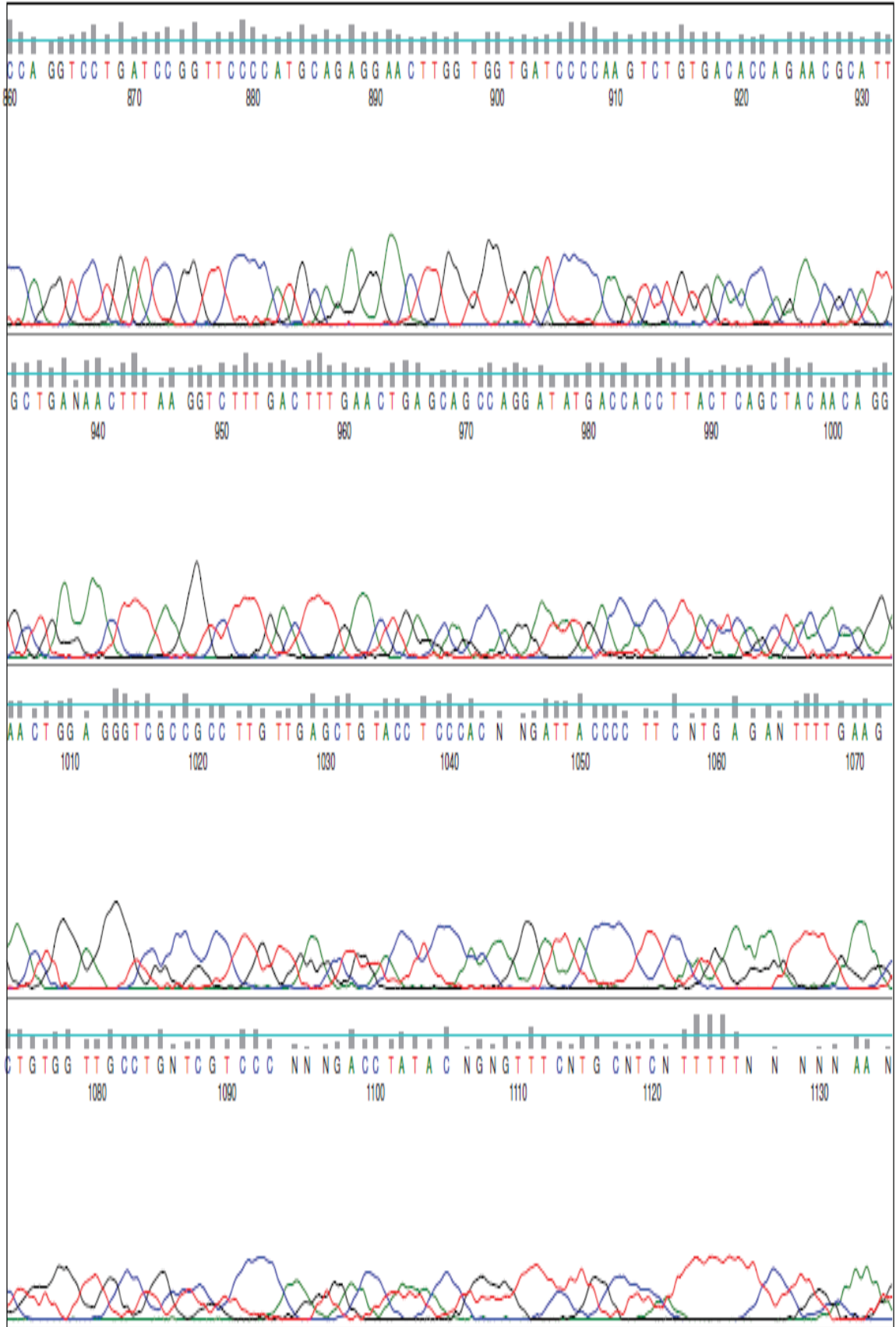
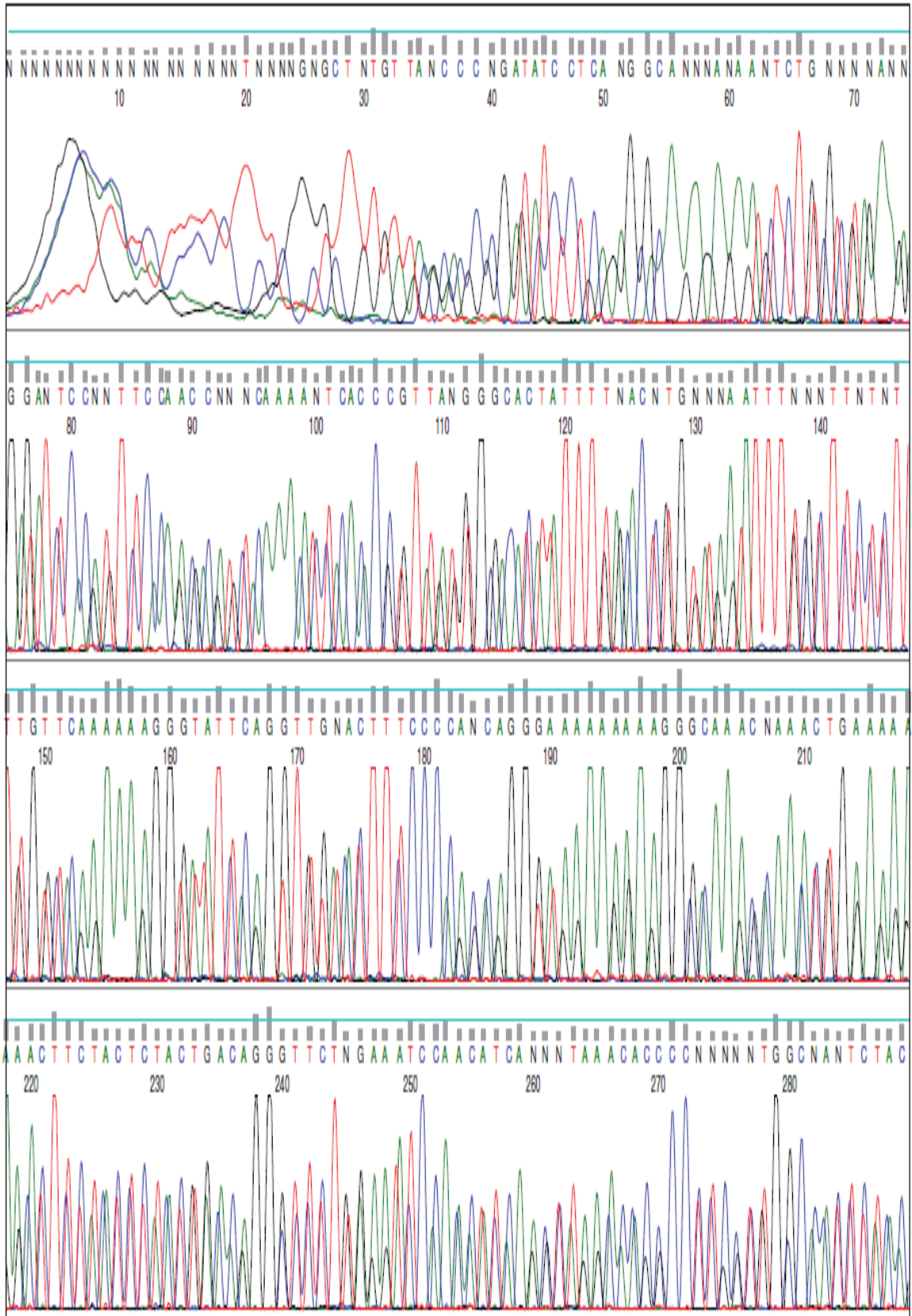


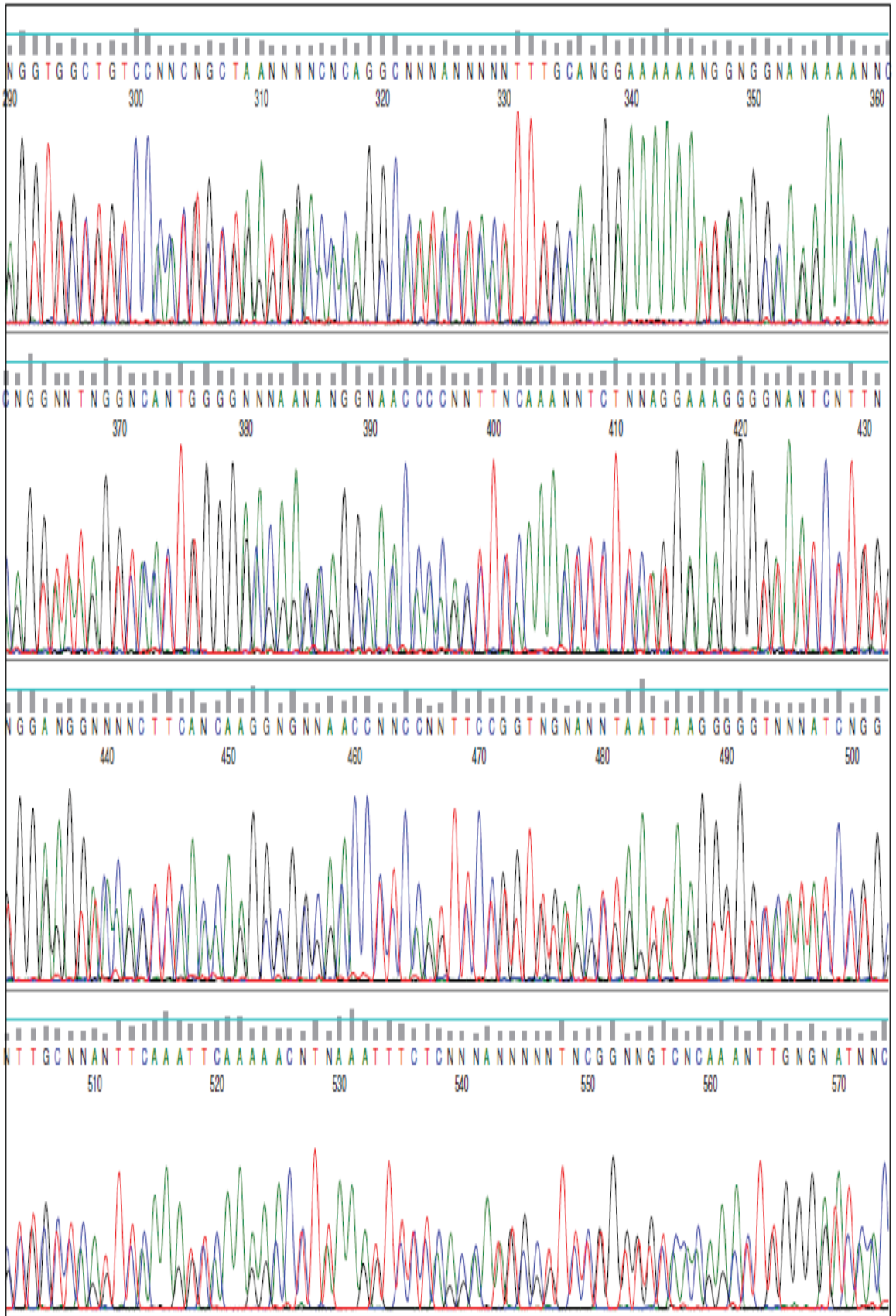


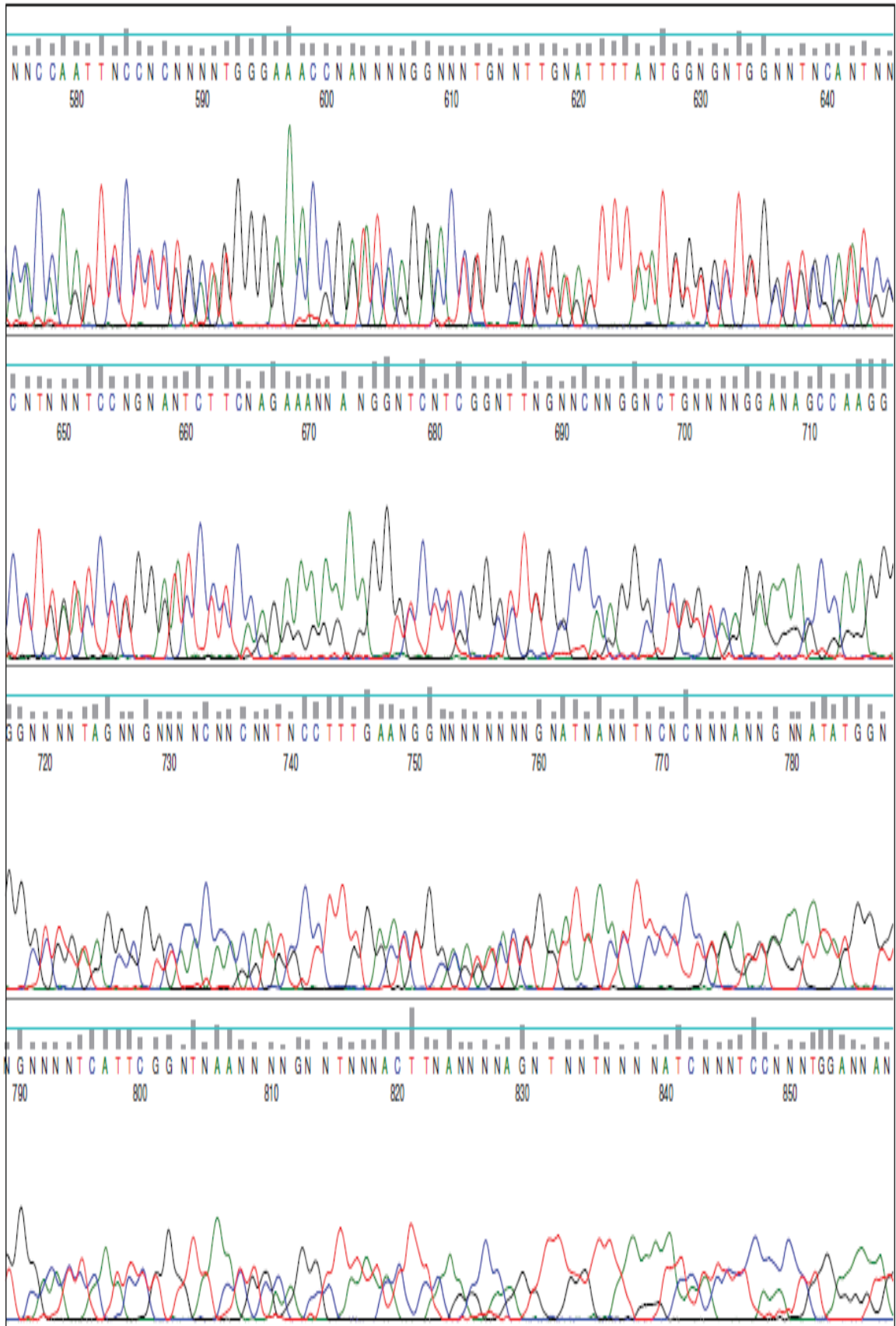
Figure 5.11: C298G Tube 1 Reverse Sequence

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AGTTCCTGTTGTAGCTGAGTAAGGTGGTCATATCCTGGCTGCTCAGTTCAAAG  
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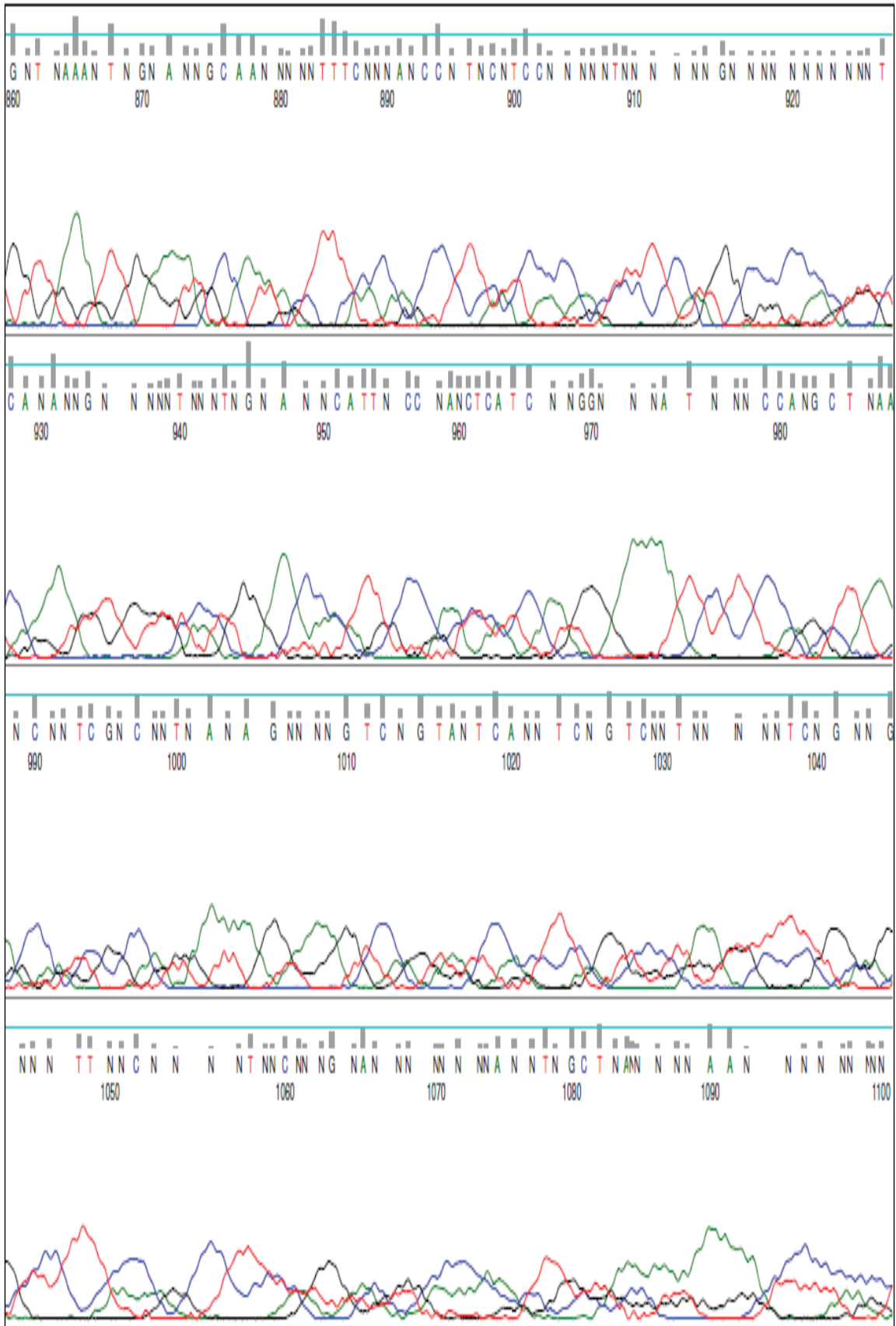
Figure 5.12: C298G Reverse DNA Chromatogram

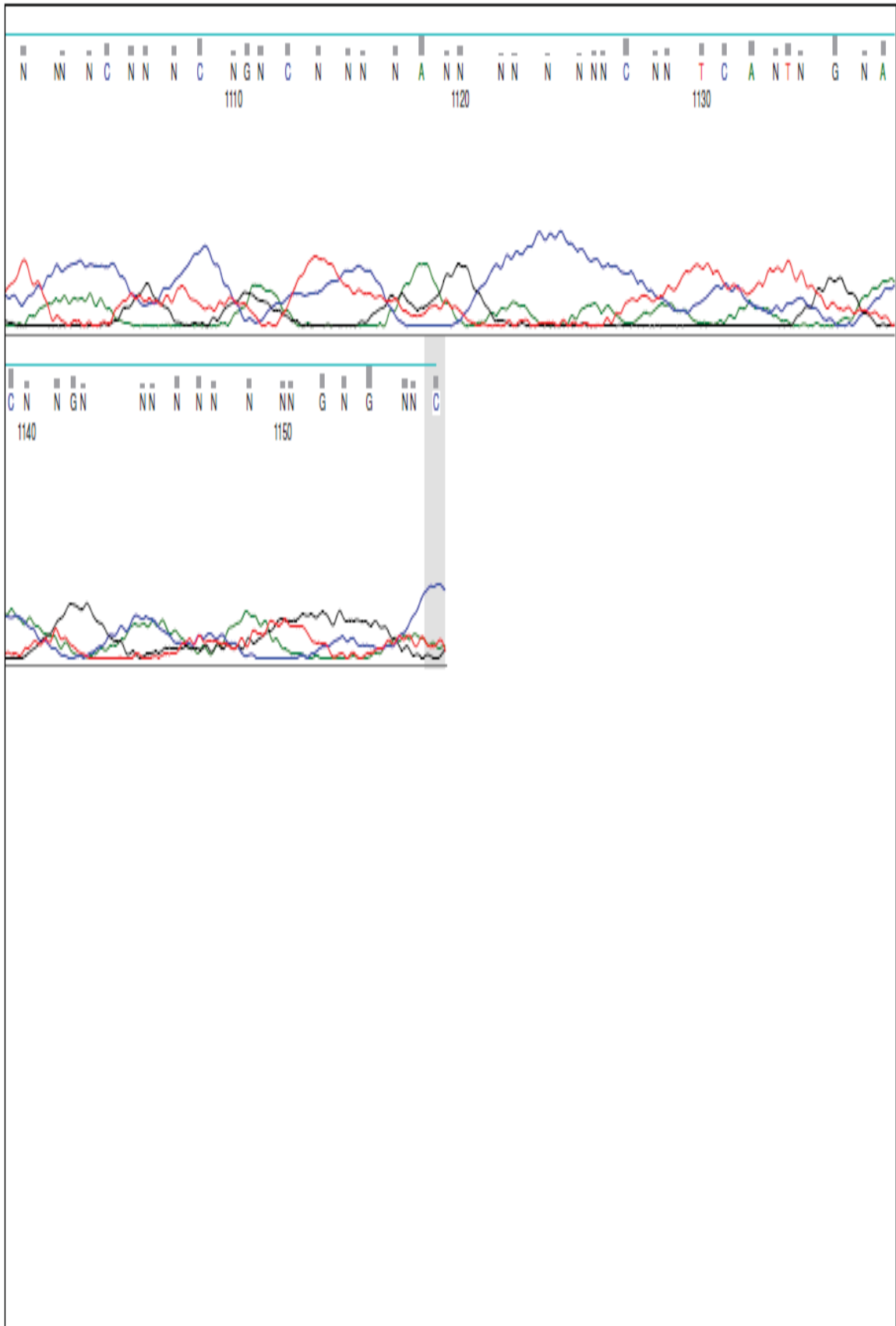










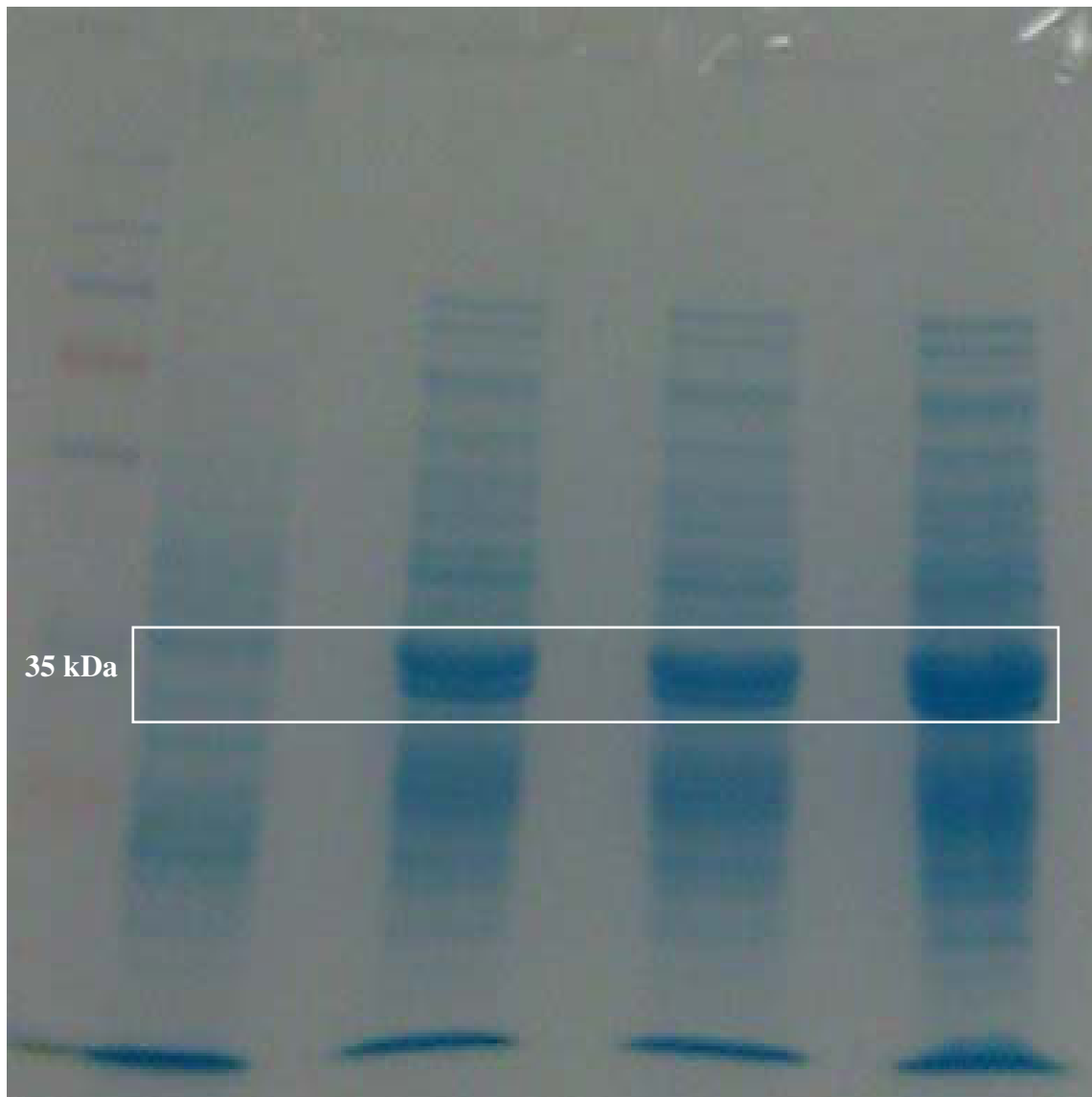


## **II. Protein Expression**

This gel (Figure 5.13) corresponds to the expression of C298G. Lanes 3-5 are the induced samples with a noticeable band at around 35 kDa. Lane 2 is the uninduced sample with no noticeable band at 35 kDa. The 36 kDa is the size of hAR so a band in the 35 kDa region confirms that aldose reductase has been expressed.

Figure 5.13: Uninduced/Induced Samples of C298G

**1**      **2**                      **3**                      **4**                      **5**

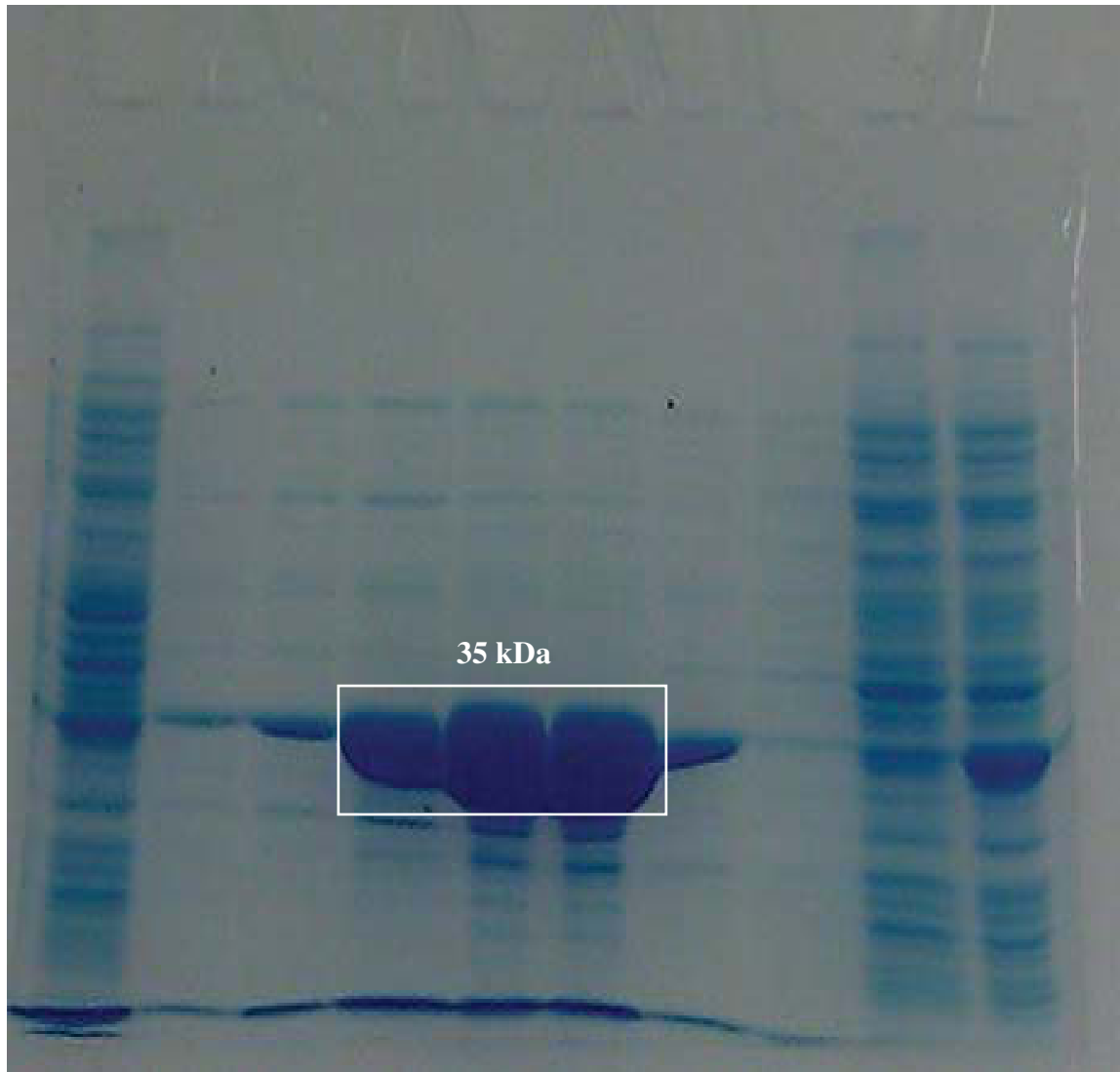


### **III. Immobilized Metal Affinity Chromatography**

This is the samples from protein purification of C298A (Figure 5.14). Lane 10 is the supernatant, lane 9 is the flow-through, lane 8 is wash 1, lane 7 is wash 2, lanes 2 -6 are the eluates from the elution buffer. Eluate 3 (lane 4), eluate 2 (lane 5) and eluate 1 (lane 6) were chosen for dialysis because they had thicker bands in the 35 kDa region.

Figure 5.14: IMAC Column Samples of C298A

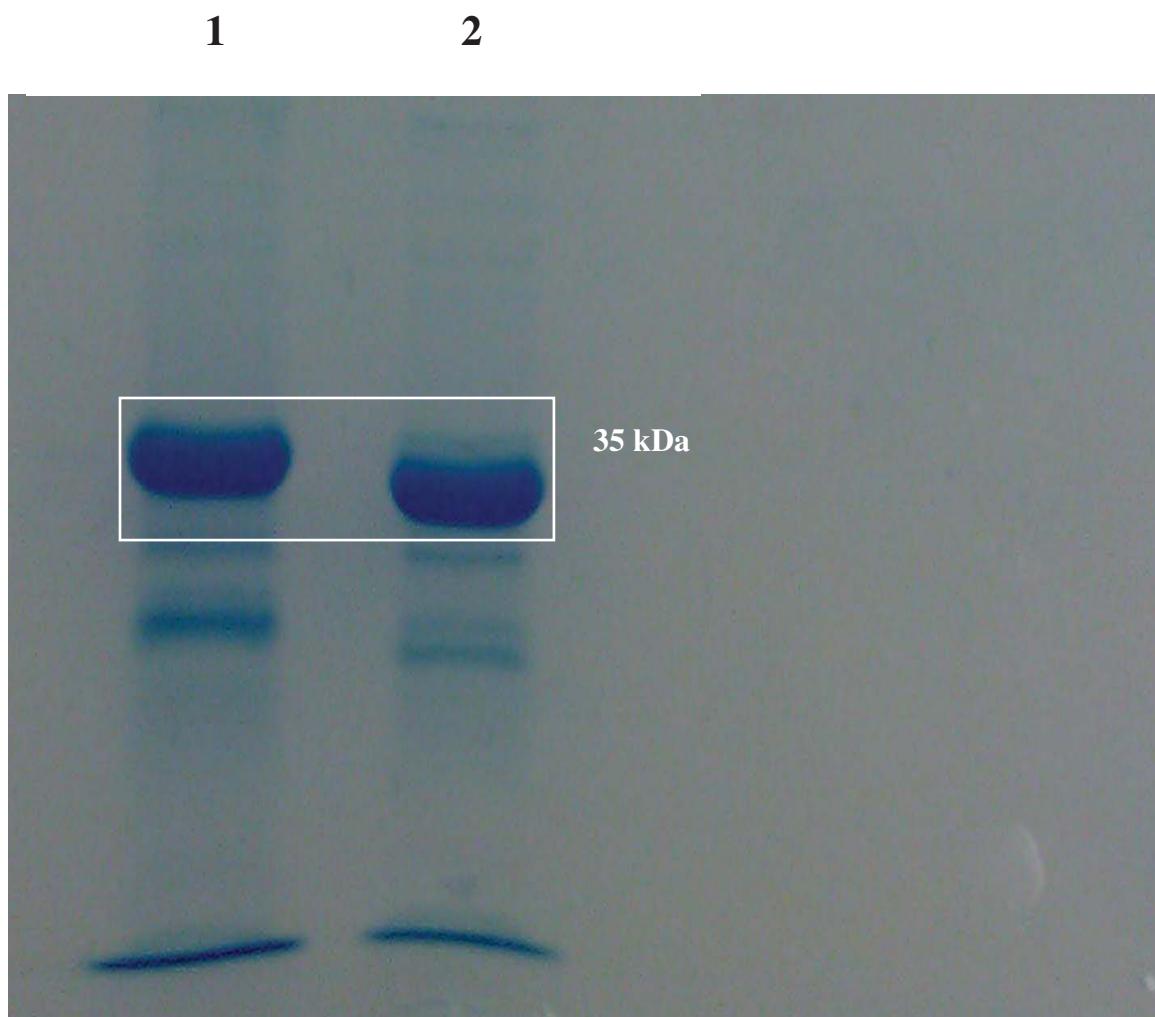
**1    2    3    4    5    6    7    8    9    10**



#### **IV. Dialysis and Thrombin Cleavage of Histidine Tag**

This is a gel from C298D (Figure 5.15) showing samples from dialysis and thrombin digestion. Lane 1 is the protein after dialysis and lane 2 is the protein after thrombin digestion. There is a slight decrease in the size of the protein band due to the cleavage of the histidine tag.

Figure 5.15: Dialysis and Thrombin Digestion Samples of C298D





## Chapter VI: Conclusion

The thesis project was a great learning experience and fortunately the mutagenesis, expression and purification of the Cys298 hAR mutants was successful. Several methods were used at each part of the project and some parts were more challenging than others due to unforeseen hurdles. Nonetheless, the set goals of this project provides the foundation for future kinetic, structure, and function studies.

During mutagenesis, primers were designed for synthesizing the mutant hAR DNA and later this DNA was amplified for transformation into *E. coli* bacteria. DNA sequencing and BioEdit software was used to analyze the quality of the mutant DNA and verify that correct mutations had been made.

Protein expression was induced with IPTG and incubation at 37°C for 4 hours. Optimal expression was achieved with 2X YT media and it was confirmed by the presence of a 36 kDa band on SDS-PAGE gels. Protein purification was carried out through sonication, immobilized metal affinity chromatography and thrombin cleavage of the histidine tag. Finally, protein quantification was completed through Bradford Reagent Assay with bovine  $\gamma$ -globulin standards. Overall, the procedures have been optimized for the reproducible production of recombinant cells of hAR mutants (C298A, C298D, C298G) for their overexpression and purification.

Future work would include measuring the Michaelis-Menten parameters of these mutants with forward reaction substrates, reverse reaction substrates, and inhibitors. Also, van't Hoff analysis could be used to study the thermodynamics of enzyme-inhibitor and enzyme-substrate interaction. Additionally, Arrhenius analysis could be performed to examine the activation energy differences between the wild-type and mutant enzymes.

The kinetic and thermodynamic properties of the C298A and C298G mutants may be similar to that of the wild-type. Both amino acids have neutral, non-polar side chains that are expected to minimally destabilize the NADPH and substrate binding sites. For the C298D mutant, the kinetic and thermodynamic properties should be more pronounced than that of the wild-type. Aspartate has an acidic, polar side chain that may significantly destabilize both binding sites.

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